Supporting Information

Electrochemical isothiocyanation of primary amines

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Material and methods

General Experimental Procedures

All reactions were carried out under aerobic conditions unless otherwise stated. All solvents and commercially available reagents were purchased from standard vendors and used without further purification unless otherwise stated. Electrolyses were performed using an IKA Electrasyn 2.0 using carbon graphite working electrode and nickel counter electrode. Analytical thin-layer chromatography (TLC) was performed using silica gel plates (0.25 mm thickness) on an aluminium support. Visualization was accomplished by irradiation with a UV lamp and/or staining with either KMnO₄ or ninhydrin. Column chromatography was performed over Silica gel 60 Å (40-63 μ mesh) using a CombiFlash Rf Lumen automatic flash chromatography system. Residual solvent was removed using a static oil pump (< 10 mbar). The cooling of reaction mixtures was achieved using an ice bath (0 °C).

NMR spectra were obtained using a JEOL ECZR 400 (¹H 399.78 MHz; ¹⁹F 376.17 MHz; ¹³C 100.53 MHz) or ECA 500 (¹H 500.16 MHz; ¹³C 125.77 MHz) spectrometer and are reported relative to the residual solvent resonances. All heteronuclear NMR spectra were ¹H-decoupled and recorded at room temperature unless otherwise stated. Data for ¹H NMR spectra are reported as follows: chemical shift (δ , ppm), coupling constant (Hz), multiplicity (s, singlet; d, doublet; t, triplet; m, multiplet; br, broad) and integration. Data for ¹³C and ¹⁹F NMR are reported in terms of chemical shift (δ , ppm). IR spectra were recorded on a Perkin Elmer Spectrum Two instrument as neat samples.

High Resolution Mass Spectrometry (HRMS) data were obtained by Dr. Iain Goodall and Dr. Perry Devo of the University of Greenwich Mass Spectrometry Service using a Waters Synapt G2 hybrid Quadrupole-orthogonal acceleration time-of-flight configuration (Waters, Manchester, UK) operating in Resolution Mode ($M/\Delta M \ge 18,000$), fitted with a Waters Acquity UPLC binary solvent chromatographic pump system. The column used was a reversed-phase Acquity BEH C18 2.1 x 50 mm, 1.7-micron bead, running a 3- minute separation with an A:B eluent mixture comprising of either deionised water with 0.1% (v:v) formic acid and acetonitrile with 0.1% (v:v) formic acid (negative mode) respectively or deionised water with 0.1% (v:v) ammonium hydroxide and acetonitrile with 0.1% (v:v) ammonium hydroxide (positive mode) respectively. Mass calibration of the instrument was performed using sodium formate cluster ions, and an orthogonal Lock-SprayTM ESI probe was used with a lock mass calibrant, leucine-enkephalin. The pseudomolecular leucine-enkephalin ion at m/z = 554.2615 (Negative Ion Mode), and m/z = 556.2771 (Positive Ion Mode), was used as the internal mass correction calibrant. Additional samples were analyzed on a Thermo LTQ Orbitrap XL coupled with a heated electrospray source (HESI). The capillary temperature was set to 275 °C and a voltage of 21 V. The sheath gas and auxiliary gas flow were set to 10 and 5 L h⁻¹ respectively and the source current and voltage set to 100 µA and 5 kV. A solution of analyte (0.1 mg/ml) and sodium formate (1% v/v) in acetonitrile was added by direct infusion (10 µL/min) into the mass spectrometer using a Hamilton syringe (250 µL).

Gas-Chromatography Mass Spectrometry (GC-MS) data were obtained using a Shimadzu Nexis GC-2030 gas chromatograph connected to a GCMS-QP2020 NX gas chromatograph mass spectrometer, equipped with an AOC-20i Plus auto injector. The column was a CD-5MS capillary column (30 m x 0.25 mm x 0.25 μ m), with helium as the carrier gas. The sample injection volume was 1 μ L, and separations run over a 5-minute period with an increasing oven temperature (gradient) between 40 – 280 °C. Results were visualised and manipulated using LabSolutions GCMS solution version 4.50.

High-Performance Liquid Chromatography-Mass Spectrometry (HPLC-MS) data were obtained using a Shimadzu LC-2050C 3D coupled with a Shimadzu LCMS-2020 FCV-20AH2. The column was an Ascentis Express 90Å AQ-C18, 2.7 µm. Results were visualised and manipulated using LabSolutions GCMS solution version

Cyclic voltammetry studies were carried out using an Autolab 302N potentiostat interfaced through Nova 2.1 software to a personal computer. Electrochemical measurements were performed in a glovebox under an atmosphere of dinitrogen with oxygen and water levels of less than 5 ppm at 298 K, with solvents that had been thoroughly degassed and purified by passing through an alumina-based purification system. Sample concentrations of 1.0 mM were used, alongside 0.1 M [$^{n}Bu_{4}N$][PF₆] supporting electrolyte concentrations. Experiments were conducted using a standard three-electrode setup comprising of a glassy carbon disc working electrode, platinum wire counter electrode, and AgCl coated silver wire as a pseudo-reference electrode. Potentials are reported relative to the [FeCp₂]^{+/0} redox couple, obtained through the addition of ferrocene to the analyte solution.

Electrochemical Reaction Setup



Figure S1: Photograph of batch electrochemical synthesis setup (disassembled)

Cyclic Voltammetry studies



Figure S2 Cyclic voltammetry on glassy carbon (diameter 3 mm) of 1 mM solution of octylamine and 0 eq (grey line), 5 eq. (black line), 20 eq (dotted line) and 50 eq (dashed line) of CS₂ in CH₂Cl₂, 0.25V.s⁻¹ scan rate.



Figure S3 Cyclic voltammetry on glassy carbon (diameter 3 mm) of a solution of neat octylamine and 5 eq. of neat CS_2 , 0.25V.s⁻¹ scan rate.

Experimental procedures A. Summary of optimisation experiments

Table S1. Optimisation of the conversion of primary alkyl amines-^a All yields displayed are isolated yields for **2a**. ^b Experiment done by adding CS_2 to an 80 mM solution of the amine.



Entry	Cathode	Anode	Equivalents	Solvent	Current	F/mol	Yield ^a	
1 ^b	C _{gr}	C _{gr}	1.5	MeOH	20	3	14	
2 ^b	C_{gr}	C_{gr}	2.5	MeOH	20	3	22	
3 ^b	C_{gr}	C_{gr}	2.5	MeOH	20	3	20	
4 ^b	C_{gr}	C_{gr}	5	MeOH	20	3	20	
5 ^b	C_{gr}	C_{gr}	10	MeOH	20	3	22	
6	Pt	Pt	2.5	MeOH	20	3	78	
7	C_{gr}	C_{gr}	2.5	MeOH	20	3	85	
8	C_{gr}	C_{gr}	2.5	EtOH	20	3	41	
9	C_{gr}	Ni	2.5	MeCN	20	3	80	
10	C_{gr}	Ni	2.5	MeOH	50	3	12	
11	C_{gr}	Ni	2.5	MeOH	10	3	82	
12	C_{gr}	Ni	2.5	MeOH	5	3	84	
13	C_{gr}	Ni	5	MeOH	5	3	90	
14	C_{gr}	Ni	5	MeOH	5	2.2	95	
15	C_{gr}	C_{gr}	5	MeOH	5	3	67	

Anode (+) Cathode (-) undivided cell CS_2 (x equiv), Base, current, *n* F/mol Solvent, rt 4a

Table S2. Optimisatior	of the conversion	of primary aryl	amines
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Entry	Cathode	Anode	Scale	Equiv.	Solvent	Base	Current	F/mol	Yield
1	C _{gr}	Ni	0.4 mmol	5	MeOH	Et₃N	20	3	68
2	C _{gr}	Ni	0.4 mmol	5	MeOH	Et₃N	10	3	63
3	C_{gr}	Ni	0.4 mmol	5	DCM/HFIP	DBU	DBU 20		52
4	C _{gr}	Ni	0.4 mmol	5	DCM/HFIP	DBU	20	5	37
5	C _{gr}	Ni	0.4 mmol	5	MeOH	-	5	3	30
6	C _{gr}	Ni	0.4 mmol	5	MeOH	Et₃N	5	3	48
7	Cgr	Ni	0.4 mmol	5	MeOH	Et₃N	5	2.2	20
8	C_{gr}	Ni	0.4 mmol	5	DCM	DBU	5	2.2	50
9	C_{gr}	Ni	0.4 mmol	5	DCM	Et_3N	5	2.2	45
10	C_{gr}	Ni	0.4 mmol	5	H ₂ O	K_2CO_3	K ₂ CO ₃ 5		0
11	C_{gr}	Ni	0.4 mmol	10	H ₂ O	K_2CO_3	5	2.2	0
12	C _{gr}	Ni	0.4 mmol	10	DCM	DBU	5	2.2	38
13	Cgr	Ni	0.4 mmol	10	DCM	Et₃N	5	2.2	48
14	C_{gr}	Ni	0.4 mmol	10	DCM/HFIP	DBU	5	2.2	52
15	C_{gr}	Ni	0.4 mmol	10	DCM/HFIP	Et_3N	5	2.2	22
16	C_{gr}	Ni	1 mmol	5	DCM 10ml	Et_3N	5	2.2	64
17	C _{gr}	Ni	1 mmol	5	DCM 10ml	DBU	5	2.2	67
18	C _{gr}	Ni	1 mmol	5	DCM	DBU	5	2.2	71

B. Synthesis of mono Boc-protected diamines

$$H_2N$$
 $H_2 + Boc_2O$ $H_2CI_2, RT, o.n$ H_2N H_2N H_2O

Synthesised according to literature procedure¹. Di-*tert*-butyl dicarbonate (0.500 g, 2.29 mmol) in CH₂Cl₂ was added dropwise over a 2 h period to a 0.25 M solution of hexamethylenediamine (1.33 g, 11.5 mmol) in CH₂Cl₂ cooled with an ice bath. The reaction mixture was stirred overnight at room temperature and filtered. The filtrate was concentrated under vaccum and the resulting oil dissolved in EtOAc was washed with brine (3 x 30 mL), dried with sodium sulfate and concentrated under vaccum. The crude product was purified by flash chromatography (100% CH₂Cl₂ \rightarrow 10% MeOH) and afforded the mono-protected diamine as a yellow liquid (0.490 g, 99%). Spectroscopic data matches with literature reports.¹



Synthesised according to literature procedure¹. Di-*tert*-butyl dicarbonate (0.500 g, 2.29 mmol) in CH₂Cl₂ was added dropwise over a 2 h period to a 0.25 M solution of trans-1,2-diaminocyclohexane (1.31 g, 11.5 mmol) in CH₂Cl₂ cooled with an ice bath. The reaction mixture was stirred overnight at room temperature and filtered. The filtrate was concentrated under vaccum and the resulting oil dissolved in EtOAc was washed with brine (3 x 30 mL), dried with sodium sulfate and concentrated under vaccum. The crude product was purified by flash chromatography (100% CH₂Cl₂ \rightarrow 10% MeOH) and afforded the mono-protected diamine as a white solid (0.487 g, 99%). Spectroscopic data matches with literature reports.²

C. Synthesis of 2-((tert-butyldimethylsilyl)oxy)ethan-1-amine



Synthesised according to literature procedure.³ To *tert*-butylchlorodimethylsilane (TBSCl, 2.71g, 18 mmol) in CH_2Cl_2 (5 mL) was added to a mixture of ethanolamine (1.00g, 16.4 mmol) and imidazole (2.23g, 32.7 mmol) in CH_2Cl_2 (33 mL) in room temperature. The mixture was reacted for 3h and then poured into water (60 mL) and extracted with DCM (3 x 30 mL). The combined organic layers were washed with brine, dried over sodium sulfate, filtered and the filtrate was concentrated under reduced pressure. The residue was then brought to high vaccum affording 2-((*tert*-butyldimethylsilyl)oxy)ethan-1-amine as a yellow liquid (2.35g, 82%). Spectroscopic data matches with literature reports.⁴

D. Electrochemical conversion of primary amines to isothiocyanates

General procedure A. Conversion of primary alkyl amines to alkyl isothiocyanate



A 5 mL IKA Electrasyn electrochemical cell was charged with the alkyl amine (0.40 mmol). Carbon disulfide (CS_2 , 2 mmol) was added then the mixture was dissolved in MeOH (5 mL). The mixture was then electrolysed at a constant current of 5 mA for 2.2F/mol. After reaching completion, the reaction mixture was poured in aqueous HCl (1.0M, 30 mL) and extracted with hexane (3 x 20 mL). The

combined organic layers were dried over MgSO₄, filtered and the solvent was removed under reduced pressure affording the clean alkyl isothiocyanate.

Synthesised according to general procedure A from benzylamine (0.0429 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2a** (0.0567 g, 95%) as a yellow oil. Isothiocyanate **2a**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 7.42-7.31 (m, 5H), 4.72 (s, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 134.0, 131.9, 128.8, 128.2, 126.7, 48.5; IR (u, cm⁻¹, neat): 2063 [N=C=S] ; HRMS (ESI) m/z: [M-H]⁻ calcd for C₈H₆NS: 148.0226; found: 148.0245



NCS

MeO Synthesised according to general procedure A from 4-methoxybenzylamine (0.0549 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2b** (0.0695 g, 97%) as a yellow oil. Isothiocyanate **2b**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 7.24-7.23 (m, 2H), 6.92-6.89 (m, 2H), 4.63 (s, 2H), 3.81 (s, 3H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 159.7, 131.9, 128.5, 126.4, 114.4, 55.4, 48.3; IR (ν , cm⁻¹, neat): 2074 [N=C=S] ; HRMS (ESI) m/z: [M-H]⁻ calcd for C₉H₈NOS: 178.0332; found: 178.0327



Br Synthesised according to general procedure A from 4-bromobenzylamine (0.0744 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2c** (0.0867 g, 95%) as a yellow oil. Isothiocyanate **2c**: ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 7.53-7.50 (m, 2H), 7.21-7.17 (m, 2H), 4.67 (s, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): $\delta_{\rm C}$ 133.4, 133.1, 132.2, 128.6, 122.5, 48.2; IR (υ, cm⁻¹, neat): 2074 [N=C=S] ; HRMS (ESI) m/z: [M-H]⁻ calcd for C₈H₅BrNS: 225.9332; found: 225.9326



F₃C Synthesised according to general procedure A from 4-(trifluoromethyl) benzylamine (0.0701 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2d** (0.0851 g, 98%) as a yellow oil. Isothiocyanate **2d**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 7.67-7.65 (d, J = 8.11 Hz, 2H), 7.46-7.44 (d, J = 8.11 Hz, 2H), 4.80 (s, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 138.3, 133.8, 130.8, 127.2, 126.0, 124.0, 48.3; ¹⁹F NMR (376 MHz, CDCl₃): δ_{F} -62.53; IR (ν , cm⁻¹, neat): 1322 [C-F], 2076 [N=C=S] ; HRMS (ESI) m/z: [M+NH₄]⁺ calcd for C9H13N₂OS: 235.0511; found: 235.0517

NCS

Synthesised according to general procedure A from 1-phenylethylamine (0.0484 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2e** (0.0620 g, 95%) as a yellow oil. Isothiocyanate **2e**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 7.42-7.32 (m, 5H), 4.95-4.90 (q, J = 6.79 Hz, 1H), 1.69-1.67 (d, J = 6.87 Hz, 3H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 140.2, 132.2, 129.0, 128.3, 125.5, 57.1, 25.1; IR (u, cm⁻¹, neat): 2081 [N=C=S] ; HRMS (ESI) m/z: [M-H]⁻ calcd for C₉H₈NS: 162.0383; found: 162.0384



NCS

Synthesised according to general procedure A from L-phenylalanine methyl ester (0.0717 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2f** (0.06904 g, 78%) as a yellow oil. Isothiocyanate **2f**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 7.37-7.30 (m, 3H), 7.24-7.21 (dd, J = 8.23, 1.82 Hz, 2H), 4.50-4.47 (dd, J = 8.42, 4,71 Hz, 1H), 3.79 (s, 3H), 3.28-3.23 (dd, J = 13.79, 4.64 Hz, 1H), 3.16-3.10 (dd, J = 13.97, 8.35 Hz, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 168.5, 137.9, 135.1, 129.4, 128.9, 127.8, 60.9, 53.3, 39.8; IR (u, cm⁻¹, neat): 1744 [C=O], 2050 [N=C=S] ; HRMS (ESI) m/z: [M+H]⁺ calcd for C₁₁H₁₂NO₂S: 222.0583; found: 222.0586

Synthesised according to general procedure A from 2-thiophenemethylamine (0.0453 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2g** (0.0192 g, 0.124 mmol, 31%) as a brown solid. Isothiocyanate **2g**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 7.30-7.29 (dd, J = 5.18 ,1.33 Hz, 1H), 7.05-7.03 (m, 1H), 6.99-6.97 (m, 1H) 4.84 (s, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{c} 136.6, 134.3, 127.2, 126.8, 126.4, 43.9; IR (u, cm⁻¹, neat): 2069 [N=C=S] ; HRMS (ESI) m/z: [M+H]⁺ calcd for C₆H₆NS₂: 155.9936; found: 155.9926

NCS Synthesised according to general procedure A from adamantanylamine (0.0605 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2h** (0.0472 g, 78%) as a yellow oil. Isothiocyanate **2h**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 2.10-2.09 (m, 3H), 1.97-1.92 (m, 6H), 1.67-1.61 (m, 6H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 129.4, 58.5, 43.8, 35.6, 29.3; IR (u, cm⁻¹, neat): 2056 [N=C=S] ; HRMS (ESI) m/z: [M+H]⁺ calcd for C₁₁H₁₆NS: 193.0920; found: 193.0924

NCS

Synthesised according to general procedure A from cyclohexylamine (0.0397 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2i** (0.0365 g, 92%) as a yellow oil. Isothiocyanate **2i**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 3.71-3.65 (m, 1H), 1.90-1.86 (m, 2H), 1.72-1.59 (m, 4H), 1.51-1.45 (m, 1H), 1.39-1.33 (m, 3H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 129.6, 55.5, 33.3, 25.1, 23.3; IR (u, cm⁻¹, neat): 2093 [N=C=S] ; HRMS (ESI) m/z: [M+H]⁻ calcd for C₇H₁₂NS: 142.0685; found: 142.0695



Synthesised according to general procedure A from *tert*-butyl((1*S*,2*S*)-2aminocyclohexyl) carbamate (0.0857 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2j** (0.0301 g, 29%) as a white solid. Isothiocyanate **2j**: ¹H NMR (500 MHz, CDCl₃): δ_{H} 4.58 (s, 1H), 3.52 (d, J = 32Hz, 2H), 2.18 – 1.93 (m, 2H), 1.76 – 1.64 (m, 2H), 1.59 (m, 1H), 1.46 (s, 9H), 1.42 – 1.18 (m, 4H) ; ¹³C {¹H} NMR (126 MHz, CDCl₃): δ_{C} 155.2, 132.4, 80.2, 60.5, 53.9, 32.4, 31.6, 28.6, 24.1, 23.6 ; IR (u, cm⁻¹, neat): 1678 [C=O], 2098 [N=C=S] ; Spectroscopic data matches with literature reports.⁵

NCS Synthesised according to general procedure A from octylamine (0.0517 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2k** (0.0512 g, 99%) as a yellow oil. Isothiocyanate **2k**: ¹H NMR (500 MHz, CDCl₃): δ_{H} 3.51-3.48 (t, J = 7.12 Hz, 2H), 1.70-1.65 (m, 2H), 1.41-1.36 (m, 2H), 1.29-1.25 (m, 8H), 0.88-0.85 (t, J = 6.98 Hz, 3H); ¹³C {¹H} NMR (126 MHz, CDCl₃): δ_{C} 129.5, 45.1, 31.8, 30.0, 29.1, 28.8, 26.6, 22.7, 14.2; IR (ν , cm⁻¹, neat): Spectroscopic data matches with literature report.⁶ ; HRMS (ESI) m/z: [M+H]⁺ calcd for C₉H₁₈NS: 172.1155; found: 172.1155

SCN Synthesised according to general procedure A with 10 equiv. CS₂ and electrolysed with a constant current of 5 mA for 4.4F/mol from hexamethylenediamine (0.0465 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2I** (0.0330 g, 71%) as a yellow oil. Isothiocyanate **2I**: ¹H NMR (500 MHz, CDCl₃): δ_{H} 3.55-3.31 (t, J = 6.51 Hz, 2H), 1.76-1.68 (m, 2H), 1.47-1.44 (m, 2H); ¹³C {¹H} NMR (126 MHz, CDCl₃): δ_{C} 130.0, 45.0, 29.8, 26.0; IR (u, cm⁻¹, neat): 2081 [N=C=S]; HRMS (ESI) m/z: [M+H]⁺ calcd for C₈H₁₃N₂S₂: 201.0515; found: 201.0522



Synthesised according to general procedure A from tert-butyl (6aminohexyl) carbamate (0.0865 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate 2m (0.0548 g, 53%) as a yellow oil. Isothiocyanate **2m**: ¹H NMR (500 MHz, CDCl₃): δ_H 4.53 (s, 1H), 3.50 (t, *J* = 6.6 Hz, 2H), 3.11 (m, 2H), 1.74 – 1.62 (m, 2H), 1.53 – 1.38 (m, 14H), 1.37 – 1.29 (m, 2H) ; ¹³C {¹H} NMR $(126 \text{ MHz}, \text{CDCl}_3)$: δ_c 156.1, 129.8, 79.3, 45.1, 40.5, 30.1, 30.0, 28.5, 26.4, 26.1; IR (ν , cm⁻¹, neat): 1689 [C=O], 2093 [N=C=S] ; HRMS (ESI) m/z [M+H]⁺ calcd for C₁₂H₂₃N₂O₂S: 259.1480; found: 259.1480

NCS HO

Synthesised according to general procedure A from 6-amino-1-hexanol (0.0469 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate 2n (0.0347 g, 74%) as a yellow oil. Isothiocyanate **2n**: ¹H NMR (500 MHz, CDCl₃): δ_H 3.65-3.62 (t, J = 6.58 Hz, 2H), 3.53-3.49 (t, J = 6.58 Hz, 2H), 1.72-1.68 (m, 3H), 1.59-1.55 (m, 2H), 1.44-1.38 (m, 4H); ¹³C {¹H} NMR (126 MHz, CDCl₃): δ_C 129.5, 62.7, 45.0, 32.5, 30.0, 26.4, 25.1; IR (u, cm⁻¹, neat): 2087 [N=C=S], 3332 [O-H]; HRMS (ESI) m/z: [M+H]⁺ calcd for C₇H₁₄NOS: 160.0791; found: 160.0784

NCS TBSO

Synthesised according to general procedure A from 2-((tertbutyldimethylsilyl)oxy)ethan-1-amine (0.0701 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2o** (0.0537 g, 62%) as a colourless solid. Isothiocyanate **2o**: ¹H NMR (500 MHz, CDCl₃): $\delta_{\rm H}$ 3.80 (t, J = 5.4 Hz, 2H), 3.57 (t, J = 5.4 Hz, 2H), 0.91 (s, 9H), 0.10 (s, 6H); ¹³C {¹H} NMR (126 MHz, CDCl₃): δ_C 132.4, 61.8, 47.8, 25.8, 18.3, -5.3; IR (υ, cm⁻¹, neat): 2087 [N=C=S] ; HRMS (ESI) m/z: [M+H]⁺ calcd for C₉H₂₀NOSSi: 218.1035; found: 218.1049



Synthesised according to general procedure A from 6-aminohexanoic acid (0.0525 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate 2p (0.0686 g, 99%) as a yellow oil. Isothiocyanate **2p**: ¹H NMR (400 MHz, CDCl₃): δ_H 3.54-3.51 (t, J = 7.40 Hz, 2H), 2.40-2.37 (t. J = 7.40 Hz, 2H), 1.75-1.64 (m, 4H), 1.51-1.44 (m, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_c 179.7, 130.3, 45.0, 33.8, 29.8, 26.1, 23.9; IR (u, cm⁻¹, neat): 1705 [C=O], 2090 [N=C=S] ; HRMS (ESI) m/z: [M-H]⁺ calcd for C₇H₁₀NO₂S: 172.0427; found: 172.0420

NCS Synthesised according to general procedure A from 1-amino-3,3diethoxypropane (0.0589 g, 0.40 mmol). Evaporation of solvent afforded isothiocyanate **2q** (0.0300 g, 40%) as a dark green oil. Isothiocyanate **2q**: ¹H NMR (500 MHz , CDCl₃): δ_{H} 4.60 (m, 1H), 3.71 – 3.57 (m, 4H), 3.50 (m, 2H), 1.96 (m, 2H), 1.20 (m, 6H) ; ¹³C {¹H} NMR (126 MHz , CDCl₃): δ_{C} 130.3, 100.2, 62.4, 41.3, 34.2, 15.4; IR (ν , cm⁻¹, neat): 2093 [N=C=S] Spectroscopic data matches with literature reports.⁷

General procedure B. Conversion of primary aryl amines to aryl isothiocyanate



A 5 mL IKA Electrasyn electrochemical cell was charged with the aryl amine (1 mmol). Carbon disulfide (CS₂, 2 mmol) and DBU (2 mmol) was added, the mixture was stirred for 5 min before being dissolved in DCM (5 mL). The mixture was then electrolysed at a constant current of 5 mA for 2.2F/mol. After reaching completion, the reaction mixture was poured in aqueous HCl (1.0M, 30 mL) and extracted with hexane (3 x 20 mL). The combined organic layers were dried over MgSO₄, filtered and the solvent was removed under reduced pressure. Filtration over silica gel with hexane was performed when necessary. The solvent was removed under reduced pressure affording the clean aryl isothiocyanate.

Synthesised according to general procedure B from aniline (0.0372 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4a** (0.0384 g, 71%) as a yellow oil. Isothiocyanate **4a**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 7.35-7.31 (m, 2H), 7.28-7.24 (m, 1H), 7.21-7.18 (m, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 135.3, 131.2, 129.6, 127.4, 125.8; IR (u, cm⁻¹, neat): 2041 [N=C=S] ; HRMS (ESI) m/z: [M-H]⁻ calcd for C₇H₄NS: 134.0070; found: 134.0091

NCS

NCS

Synthesised according to general procedure B from 4-isopropylaniline (0.0541 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4b** (0.0674 g, 0.38 mmol, 95%) as a yellow oil. Isothiocyanate **4b**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 7.32-7.18 (m, 4H), 3.31-3.25 (m, 1H), 1.30-1.29 (d, J = 6.92 Hz, 6H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 144.7, 135.3, 129.4, 127.8, 126.9, 126.9, 126.4, 29.7, 22.9; IR (u, cm⁻¹, neat): 2077 [N=C=S] ; HRMS (ESI) m/z: [M+NH₄]⁺ calcd for C₁₀H₁₅N₂S: 195.0950; found: 195.0945

NCS

NCS

Synthesised according to general procedure B from 4-*tert*-butylaniline (0.0597 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4c** (0.0712 g, 93%) as a brown solid. Isothiocyanate **4c**: ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 7.38-7.34 (m, 2H), 7.17-7.14 (m, 2H), 1.31 (s, 9H); ¹³C {¹H} NMR (101 MHz, CDCl₃): $\delta_{\rm C}$ 150.8, 134.5, 128.4, 126.5, 125.4, 34.8, 31.3; IR (u, cm⁻¹, neat): 2050 [N=C=S] ; HRMS (ESI) m/z: [M+Na]⁺ calcd for C₁₁H₁₃NSNa: 214.0661; found: 214.0641

Synthesised according to general procedure B from 4-methoxyaniline (0.0493 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4d** (0.0529 g, 80%) as a yellow oil. Isothiocyanate **4d**: ¹H NMR (400 MHz, CDCl₃): δ_H 7.15-7.12 (m, 2H), 6.85-6.82 (m, 2H), 3.78 (s, 3H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_C 158.6, 133.8, 127.0, 123.5, 114.8, 55.6; IR (ν , cm⁻¹, neat): 2050 [N=C=S] ; HRMS (ESI) m/z: [M+MeOH+H]⁺ calcd for C₉H₁₂NO₂S: 198.0583; found: 198.0589



NCS

Synthesised according to general procedure B from 2,4,6-trimethylaniline (0.0541 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4e** (0.0581 g, 82%) as a white solid. Isothiocyanate **4e**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 6.85 (s, 1H), 2.33 (s, 6H), 2.28 (s, 3H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 137.1, 134.9, 134.9, 128.8, 126.9, 21.1, 18.6; IR (u, cm⁻¹, neat): 2105 [N=C=S] ; HRMS (ESI) m/z: [M+H]⁺ calcd for C₁₀H₁₂NS: 178.0685; found: 178.0692

CI Synthesised according to general procedure B from 4-chloroaniline (0.0510 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4f** (0.0271 g, 40%) as a yellow oil. Isothiocyanate **4f**: ¹H NMR (400 MHz, CDCl₃): δ_H 7.32-7.30 (m, 2H), 7.16-7.13 (m, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_C 136.8, 133.0, 130.0, 129.9, 127.0; IR (υ , cm⁻¹, neat): 2075 [N=C=S] ; HRMS (ESI) m/z: [M+MeO]⁻ calcd for C₈H₇OCINS: 199.9942; found: 199.9937

Br Synthesised according to general procedure B from 2-bromoaniline (0.0688 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4g** (0.0788 g, 92%) as a yellow solid. Isothiocyanate **4g**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 7.56-7.54 (dd, J = 8.16, 1.36 Hz, 1H), 7.27-7.21 (m, 2H), 7.11-7.08 (m, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 138.1, 133.5, 131.6, 128.4, 128.3, 127.2, 120.9; IR (u, cm⁻¹, neat): 2038 [N=C=S] ; HRMS (ESI) m/z: [M+H]⁺ calcd for C₇H₅BrNS: 213.9321; found: 213.9330

F Synthesised according to general procedure B from 4-fluoroaniline (0.0444 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4h** (0.0539 g, 88%) as a brown solid. Isothiocyanate **4h**: ¹H NMR (400 MHz, CDCl₃): δ_H 7.23-7.19 (m, 2H), 7.07-7.02 (m, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_C 162.5, 160.0, 135.9, 127.6, 116.9; ¹⁹F-NMR (376 MHz, CDCl₃) δ_F -111.92; IR (u, cm⁻¹, neat): 1499 [C-F], 2050 [N=C=S] ; HRMS (ESI) m/z: [M+H]⁺ calcd for C₇H₅FNS: 154.0121; found: 154.0122



NCS

NCS

Synthesised according to general procedure B from 3-trifluoromethylaniline (0.0644 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4i** (0.0512 g, 63%) as a yellow solid. Isothiocyanate **4i**: ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 7.54-7.47 (m, 3H), 7.41-7.39 (m, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): $\delta_{\rm C}$ 138.4, 132.6, 132.5, 130.4, 129.0, 123.9, 123.4, 122.8; ¹⁹F-NMR (376 MHz, CDCl₃) $\delta_{\rm F}$ -62.89; IR (u, cm⁻¹, neat): 1329 [C-F], 2049 [N=C=S] ; HRMS (ESI) m/z: [M+H]⁻ calcd for C₈H₅F₃NS: 204.0100; found: 204.0101



HO Synthesised according to general procedure B from 2-(4-aminophenyl) acetic acid (0.0605 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4j** (0.0641 g, 83%) as a yellow oil. Isothiocyanate **4j**: ¹H NMR (400 MHz, CDCl₃): δ_H 7.28-7.25 (m, 2H), 7.20-7.18 (m, 2H), 3.65 (s, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_C 177.1, 136.0, 132.6, 130.8, 130.7, 126.1, 40.6; IR (u, cm⁻¹, neat): 1692 [C=O], 2093 [N=C=S] ; HRMS (ESI) m/z: [M-H]⁺ calcd for C₉H₆NO₂S: 192.0114; found: 192.0107



O Synthesised according to general procedure B from methyl-2-aminobenzoate (0.0605 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4k** (0.0525 g, 68%) as a yellow solid. Isothiocyanate **4k**: ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 7.99-7.97 (dd, J = 7.91, 1.55 Hz, 1H), 7.52-7.49 (m, 1H), 7.35-7.28 (m, 2H), 3.97 (s, 3H); ¹³C {¹H} NMR (101 MHz, CDCl₃): $\delta_{\rm C}$ 165.2, 135.9, 133.4, 131.9, 130.6, 127.8, 127.0, 126.5, 52.9; IR (u, cm⁻¹, neat): 1719 [C=O], 2089 [N=C=S] ; HRMS (ESI) m/z: [M+H]⁺ calcd for C₉H₈NO₂S: 194.0270; found: 194.0269



Synthesised according to general procedure B from 3-ethynylaniline (0.0469 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **4I** (0.0280 g, 44%) as a yellow oil. Isothiocyanate **4I**: ¹H NMR (400 MHz, CDCl₃): δ_H 7.39-7.36 (m, 1H), 7.33-7.32 (m, 1H), 7.30-7.28 (m, 1H), 7.21-7.18 (m, 1H), 3.13 (s, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_C 136.8, 131.7, 131.0, 129.7, 129.2, 126.2, 123.9, 82.0, 79.0; IR (u, cm⁻¹, neat): 2033 [N=C=S] ; HRMS (ESI) m/z: [M-H]⁻ calcd for C₉H₄NS: 158.0070; found: 158.0069



NCS Synthesised according to general procedure B from 4-aminoindole (0.0529 g, 0.40 mmol). Purification through SiO₂ with hexane as solvent afforded isothiocyanate **30** (0.0118 g, 17%) as a yellow oil. isothiocyanate **30**: ¹H NMR (400 MHz, CDCl₃): δ_{H} 8.29 (s, 1H), 7.29-7.26 (m, 1H), 7.21-7.20 (t, J = 2.97 Hz, 1H), 7.08-7.04 (t, J = 7.96 Hz, 1H), 6.93-6.90 (m, 1H), 6.66-6.65 (m, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ_{C} 136.7, 135.9, 125.4, 125.3, 122.9, 122.3, 116.8, 111.0, 100.4; IR (υ, cm⁻¹, neat): 3419 [N-H], 2052 [N=C=S]; HRMS (ESI) m/z: [M-H]⁻ calcd for C₉H₅N₂S: 173.0179; found: 173.0173

References

- Pockes, S.; Wifling, D.; Buschauer, A.; Elz, S. Structure-Activity Relationship of Hetarylpropylguanidines Aiming at the Development of Selective Histamine Receptor Ligands⁺. *ChemistryOpen* **2019**, *8* (3), 285–297.
- (2) Monasterolo, C.; Adamo, M. F. A. Vinylogous Nitro-Haloform Reaction Enables Aromatic Amination. *Org. Lett.* **2022**, *24* (26), 4729–4733.

- (3) McDonnell, C. M.; Ghanim, M.; Mike Southern, J.; Kelly, V. P.; Connon, S. J. De-Novo Designed β-Lysine Derivatives Can Both Augment and Diminish the Proliferation Rates of E. Coli through the Action of Elongation Factor P. *Bioorg. Med. Chem. Lett.* **2022**, *59*, 128545. https://doi.org/10.1016/j.bmcl.2022.128545.
- (4) Greve, E.; Lindeman, S. V.; Scartelli, C.; Lin, L.; Flaumenhaft, R.; Dockendorff, C. Route Exploration and Synthesis of the Reported Pyridone-Based PDI Inhibitor STK076545. *Org. Biomol. Chem.* 2020, 18 (34), 6665–6681. https://doi.org/10.1039/D0OB01205J.
- (5) Smith, J.; Liras, J. L.; Schneider, S. E.; Anslyn, E. V. Solid and Solution Phase Organic Syntheses of Oligomeric Thioureas. *J. Org. Chem.* 1996, 61 (25), 8811–8818. https://doi.org/10.1021/jo9614102.
- (6) Sun, N.; Li, B.; Shao, J.; Mo, W.; Hu, B.; Shen, Z.; Hu, X. A General and Facile One-Pot Process of Isothiocyanates from Amines under Aqueous Conditions. *Beilstein J. Org. Chem.* 2012, 8 (1), 61– 70. https://doi.org/10.3762/bjoc.8.6.
- (7) Fisyuk, A. S.; Mukanov, A. Yu. Synthesis of 3-Isothiocyanatopropion-and-Butyraldehyde Diethyl Acetals and Their Reactions with n-Nucleophiles. *Russ. J. Org. Chem.* **2006**, *42* (9), 1269–1274.

NMR Spectra



































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220 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 -20 (ppm)

S80