

1           **Spent coffee ground as a renewable source of energy:**  
2                           **analysis of bulk powder flowability**

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4                           **L. M. Sousa<sup>1</sup> and M. C. Ferreira<sup>1\*</sup>**

5           <sup>1</sup>Drying Center for Pastes, Suspensions, and Seeds, Chemical Engineering Department,  
6                           Federal University of São Carlos, P.O. Box 676, 13565-905 São Carlos, Brazil

7  
8   **ABSTRACT**

9           The main use for Spent Coffee Ground (SCG) produced in the soluble coffee industry is  
10 the thermal energy generation in the industry itself. Throughout processing, SCG is submitted  
11 to unit operations strongly dependent on the powder flow behavior. In this study, we  
12 evaluated two classical flowability indexes of non-consolidated SCG powder: the angle of  
13 repose (AoR) and the Hausner ratio (HR). The influence of the mean diameter, particle size  
14 distribution and moisture content on AoR and HR of SCG was analyzed for powders with  
15 particle mean sizes from 225 to 550  $\mu\text{m}$ . Values of  $\text{HR} > 1.35$  and  $\text{AoR} > 45^\circ$ , which  
16 characterize a poor flowability, were observed for powders of mean particle size close to 350  
17  $\mu\text{m}$  and for mixtures with more than 40% of fine particles in their composition. The AoR was  
18 sensitive to powder size distribution, and powders with similar mean sizes presented higher  
19 values of AoR when the mixture size span was larger. For powders with moisture content up  
20 to 50%, the flowability indexes were not significantly affected by the moisture content. The  
21 Linear Mixture-Packing Model was used to predict the packed-bed void fraction for binary  
22 and ternary mixtures made up from the combination of three base powders of different sizes.

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\* Corresponding author. Tel.: +55 16 33518442. E-mail address: [mariaf@ufscar.br](mailto:mariaf@ufscar.br) (M. C. Ferreira).  
Abbreviations: AoR-angle of repose; HR-Hausner ratio; PSD-particle size distribution; SCG-spent coffee  
ground; SCI-soluble coffee industry.

23 The model was used to build up a ternary diagram that is able to estimate HR of the powder  
24 mixtures. Additionally, an equation was fitted to correlate HR and AoR. The proposed  
25 diagram and the fitted equation may provide insight into the flow behavior and help process  
26 design of industrial plants that handle SCG.

27 *Keywords:* spent coffee ground; biomass, flowability; Hausner ratio; Angle of Repose.

28

## 29 **1. Introduction**

30 In 2016, more than 9 million tons of coffee beans were produced worldwide  
31 (International Coffee Organization, 2017), generating a revenue around US\$ 25 billion  
32 (Brazilian Coffee Exporters Council, 2017). In emerging countries well-known as traditional  
33 coffee producers, such as Brazil, Colombia and Vietnam, coffee is a commodity that plays a  
34 major contribution in the trade market. Fruit and beans are processed to produce the coffee  
35 powder, which is further sent to the Soluble Coffee Industry (SCI) or directly for domestic  
36 brewing. These processes generate a huge amount of solid residue, the Spent Coffee Ground  
37 (SCG). Around 2.5 million tons of SCG were produced in 2016, which was readily available  
38 for industrial processing (Instant Coffee Market, 2017; Mussatto, Machado, Martins, &  
39 Teixeira, 2011). If disposed of in landfills without a previous treatment, this residue might  
40 cause serious environmental issues due to its high organic load and acidity (Mussatto et al.,  
41 2011). The high calorific power of SCG (close to 25 kJ/kg, which is similar to that of the  
42 coal) (Silva, Nebra, Machado Silva, & Sanchez, 1998), makes it an attractive biomass for  
43 thermal energy generation in SCI facilities. Applications involving the production of biodiesel  
44 (Al-Hamamre, Foerster, Hartmann, Kröger, & Kaltschmitt, 2012; Kondamudi, Mohapatra, &  
45 Misra, 2008; Liu, Tu, Knothe, & Lu, 2017), fuel pellets (Kondamudi et al., 2008; P. Li,  
46 Kanda, & Makino, 2014), hydrogen and ethanol (Mussatto et al., 2011) have been reported  
47 for this solid waste but most of it is burned in SCI furnaces for steam production. Almost the

48 total energy required in SCI is supplied by SCG (Silva et al., 1998), thus significantly  
49 reducing the soluble coffee cost production. An effective handling of biomass powders  
50 through the related unit operations is of key importance to the success of those  
51 thermochemical processes.

52 In spite of the attractive energy balance, some critical issues can arise in handling  
53 biomass powders, such as the blockage of pipes, feeding devices or silos outlets (Dai, Cui, &  
54 Grace, 2012). Such occurrences may prevent a continuous and uniform flow to furnaces,  
55 reactors and so on, thus disrupting operations and reducing the overall efficiency and  
56 productivity. The effectiveness of feeding, storage, compaction and transport operations is  
57 strongly dependent on the powder flow behavior. The ability of a powder to flow is ruled by  
58 complex interactions between attractive and external field forces. If the weight of the particles  
59 is much larger than the attractive interparticle forces they will flow well, otherwise, a  
60 cohesive behavior is expected. Interparticle adhesion is originated by intermolecular forces  
61 including van der Waals forces, local chemical bonds, electrostatic charges, and bridging  
62 forces originated by liquid surface tension (Li, Rudolph, Weigl, & Earl, 2004). Van der Waals  
63 forces of molecular origin are dominant in fine dry powders and the bridging forces become  
64 significant as the water saturation in moist bulk powders increases (Althaus & Windhab,  
65 2012; Castellanos, 2005). The numerous mechanisms that play a role in powder flowability  
66 have not been fully understood yet and quantifying flowability is quite a challenging task.  
67 However, it is known to depend on a number of factors including the particle size, size  
68 distribution, shape, surface texture, surface energy, chemical composition, moisture content,  
69 vessel geometry, packing history, among others (Li et al., 2004).

70 The Hausner ratio (HR) and the Angle of Repose (AoR) are indexes based on easily  
71 measurable properties that serve as useful flowability indicators for bulk powders in industrial  
72 and practical applications. The first one is based on the ratio of tapped to loose bulk density

73 and is considered a measure of powder cohesion, while the latter is obtained from powder  
74 flow under gravitational discharge (Abdullah & Geldart, 1999). As particle size distribution  
75 and moisture content of SCG produced in SCIs can vary in a wide range (Silva et al., 1998),  
76 assessing how SCG flowability is affected by these variables is useful for monitoring process  
77 operations and quality features. Although the composition and energetic characteristics of  
78 SCG have been widely investigated in the literature (Ballesteros, Teixeira, & Mussatto, 2014;  
79 Campos-Vega, Loarca-Piña, Vergara-Castañeda, & Oomah, 2015; Mussatto et al., 2011; Silva  
80 et al., 1998), to the best of authors' knowledge, no study was focused on assessing their basic  
81 flowability characteristics.

82 This study is aimed at evaluating flow indexes of non-consolidated SCG powders. The  
83 AoR and HR were determined for dry and moist powders in a size range from 225 to 550  $\mu\text{m}$   
84 and the influence of mean size and moisture content on the flowability indexes was evaluated.  
85 To be able to cover a wide range of particle's size and particle size distribution, three SCG  
86 powder samples made of fine, intermediary and coarse size particles were used to produce  
87 binary and ternary mixtures. A ternary diagram was built up to predict HR of SCG powder  
88 mixtures based on the Modified-Linear Mixture Packing model and an equation correlating  
89 HR and AoR was fitted based on the experimental data.

90

## 91 **2. Material and Methods**

92 The procedures used for sample preparation, particle and bulk powder characterization  
93 and measurement of flow indexes will be presented in this section.

94

### 95 **2.1 Sample preparation**

96 The SCG was obtained after brewing an ordinary Brazilian ground coffee made up from  
97 a blend of Arabica and Robusta grains (brand Três Corações), acquired in São Carlos-SP,

98 Brazil. To reduce the powder's moisture content, 5 mm thick layer samples were oven-dried  
99 at  $105\pm 2^{\circ}\text{C}$  for 24h and stored under ambient temperature until further use.

100 The particle size analysis of the original dried sample was performed by sieving it  
101 according to the ASTM standards (ASTM International, 2010), in triplicate assays. SCGs  
102 samples of 100 g were poured into a set of stainless steel sieves, with size apertures  
103 decreasing from 2360 to 63  $\mu\text{m}$ . The set of sieves was coupled to a shaker (RETSCH AS200)  
104 and vibrated during 30 min at an amplitude of 1.5 mm. The powder mass fraction retained in  
105 the sieves of size aperture above 600  $\mu\text{m}$ , which was only 20% of the total, was discarded and  
106 the remaining portion was sieved again from 600 to 150  $\mu\text{m}$ . This portion was separated into  
107 three fractions denominated as the base samples, A, B and C. Sample A is made up of  
108 particles with sizes between 600 and 500  $\mu\text{m}$  ( $d_{svA}=550 \mu\text{m}$ ), sample B contains the particles  
109 retained between 500 and 300  $\mu\text{m}$  ( $d_{svB}=400 \mu\text{m}$ ), and sample C those retained between 300  
110 and 150  $\mu\text{m}$  ( $d_{svC}=225 \mu\text{m}$ ). Additional mixtures were produced by combining mass fractions  
111 ( $y$ ) of 20, 40, 60 and 80% of the base samples, resulting in 12 binary and 6 ternary mixtures.  
112 The mixtures were homogenized prior to the assays. Tests were carried out in triplicate for the  
113 base powders, binary and ternary mixtures, therefore 63 dry samples were prepared and  
114 analyzed.

115

## 116 2.2 Powder characterization

117 The samples were characterized according to the methods described in the following  
118 sections.

119

### 120 2.2.1 Mean diameter

121 The mean diameter of the base samples, binary and ternary mixtures were calculated  
122 from Eq. (1), in which  $y_i$  is the mass fraction of the base sample (A, B or C) in the mixture  
123 and  $d_{svi}$  is the sieve mean diameter of the base sample.

124

$$d_S = 1/\sum(y_i/d_{svi}) \quad (1)$$

125

#### 126 2.2.2 Initial moisture content

127 The initial moisture content was determined by the gravimetric method, in triplicate.  
128 Powder samples of 3 g were kept in an oven (FANEM, model 315SE) at 105 °C for 24 h.

129

#### 130 2.2.3 Solid and particle densities

131 The solid density ( $\rho_s$ ) was determined for the SCG original dry sample using a helium  
132 gas pycnometer (Ultrapycnometer 1000, Quantachrome Instruments), in triplicate. The  
133 particle density ( $\rho_p$ ) was determined for the dry base powders A, B and C using a liquid  
134 pycnometer, with a replicate. The pycnometer has a volume of 25 cm<sup>3</sup> and the liquid used was  
135 methanol.

136

#### 137 2.2.4 Particle morphology

138 Particles shape and surface texture were evaluated through a scanning electron  
139 microscope (SEM). The SCG base samples were glued over a carbon tape and coated with  
140 gold. The analysis was performed using a microscope (FEI Inspect S50) under a high vacuum  
141 atmosphere.

142

#### 143 2.2.4 Loose and tapped bulk densities

144 The loose and tapped bulk densities were obtained according to the procedure described  
145 elsewhere (Organization, 2012), in triplicate. Powder samples (50-70g) were poured through a  
146 funnel into a 250 cm<sup>3</sup> and 2.2 cm diameter graduated glass cylinder. In all the assays, the  
147 powder filled more than 60% of the cylinder volume. The vessel was placed into a proper  
148 tapping device that allowed the vessel to be successively lifted at a height of 3 cm and  
149 released (Campos & Ferreira, 2013). The powder volume was periodically recorded from N=0  
150 up to N=1250 taps, when no more variation in the powder volume was observed. The assays  
151 were performed at ambient temperature and relative humidity between 45 and 70%, which is  
152 sufficient to eliminate static charges. Both the loose (initial) and tapped (final) bulk densities  
153 were obtained by dividing the powder mass in the cylinder by its respective volume. The  
154 loose or tapped void fractions were determined according to:

$$\varepsilon = 1 - (\rho_b^* / \rho_p) \quad (2)$$

156 where  $\rho_b^*$  is the loose ( $\rho_{lb}$ ) or tapped ( $\rho_{tb}$ ) bulk densities and  $\rho_p$  is the particle density. It is  
157 worth noting that Eq. (2) applies to estimate the void fractions of dry samples.

158

#### 159 2.2.6 Powder humidification

160 To vary the powder moisture content, a humidification procedure was performed. A  
161 powder sample of 50 g was mixed with water in glass flasks, which were sealed and stored at  
162 4°C for 60 h. After every 24 h, the flasks were opened and the powders were homogenized to  
163 ensure uniform moisture distribution.

164 The water amount added to the flasks were determined based on the established  
165 moisture content for a sample and covered a range of moisture content up to 50% (wet basis).  
166 For each base sample, four different moisture contents were analyzed in triplicate, therefore a

167 set of 36 moist samples was prepared. The moisture content in wet basis and the water  
168 saturation stage (S) in the bulk powders were calculated by:

169

$$MC = m_w / (m_w + m_s) \quad (3)$$

$$S (\%) = V_w / (V_t - V_s) \quad (4)$$

170 where  $m_w$  is the mass of liquid water in the packed-bed,  $m_s$  is the mass of solids,  $V_w$  is the  
171 total volume of liquid water in the packed-bed;  $V_t$  is the total volume of the compacted dry  
172 bed, and  $V_s$  is the volume occupied by the solids (based on  $\rho_s$ ).

173

## 174 2.3 Flowability indicators

175 The powder flowability was assessed through measurements of AoR and HR, as  
176 described in the following sections. The criteria for flowability classification based on these  
177 indexes are described in details elsewhere (Tan, Morton, & Larson, 2015). According to these  
178 criteria, the transition from a *passable* to a *poor* flowability, which defines the likely limit to  
179 flow issues occurs for AoR and HR higher than  $45^\circ$  and 1.35, respectively.

180

### 181 2.3.1 Hausner ratio (HR)

182 HR was calculated from the ratio of the tapped to the loose bulk density, according to:

183

$$HR = \rho_{tb} / \rho_{lb} \quad (5)$$

184

### 185 2.3.2 Angle of Repose (AoR)

186 The angle of repose was determined by the funnel method, according to (ASTM  
187 International, 2000). The powders were poured into the funnel and discharged by gravity over  
188 a white paper surface, forming a conical heap of height H. The conical bed circumference was

189 drawn with a pen and its diameter (D) was taken as the average of four records. The  
190 experiments were performed in triplicate and AoR was calculated by:

191

$$AoR = \tan^{-1}(2H/(D - d)) \quad (6)$$

192

## 193 2.4 Statistical analysis

194 The one-way analysis of variance (ANOVA) based on the Tukey test was used to verify  
195 whether there was any statistically significant difference ( $p < 0.05$ ) between the means of the  
196 different groups of bulk densities and flow properties. The data are reported as mean values  
197 and their standard deviations.

198 The empirical equations fitted to the experimental data using the Excel software  
199 (Microsoft) and the determination coefficients for the equations ( $R^2$ ) are reported.

200

## 201 3. Results and Discussion

202 The results for powder characterization and measurement of flowability indexes are  
203 presented in the following sections.

204

### 205 3.1 Characterization of SCG base dry samples

206 The main physical properties of the base samples are summarized in Table 1. The  
207 particle densities ( $\rho_p$ ) of the base samples are similar to the value reported by Silva et al.  
208 (1998) for SCGs ( $1200 \text{ kg/m}^3$ ) (Silva et al., 1998). The solid density is equal to  $1315 \pm 3 \text{ kg/m}^3$ ,  
209 which is similar to values reported for fresh coffee powders from Colombia and Mexico  
210 (respectively  $1361$  and  $1355 \text{ kg/m}^3$ ) (Singh, Singh, Bhamidipati, Singh, & Barone, 1997).  
211 Based on their mean size and particle density, the base powders A and B are classified as  
212 *Geldart B* and powder C is classified as *Geldart A* (Geldart, 1973). According to this

213 classification, samples A and B are coarse powders and sample C is categorized as a fine  
214 powder, but not a cohesive type.

215 (Table 1)

216 The loose bulk densities of powders A and B are higher than those of sawdust, rice  
217 straw (Guo, Chen, Xu, & Liu, 2015), sugarcane bagasse and corn stover (Dhiman et al.,  
218 2016). Due to the high combustion heat value of SCG powders, this is an appealing result  
219 concerning economic feasibility of using SCG as a fuel feedstock, as the costs for storage and  
220 transport would be lower and its energy density (in MJ/m<sup>3</sup>) higher in comparison to those  
221 biomasses. After drying, the maximum moisture content of the base samples is equal to 5.4%  
222 (w.b.), and the saturation level (S) is less than 4%, characteristic of a pendular state in which  
223 discrete liquid bridges bind the particles together (Althaus & Windhab, 2012). Therefore, the  
224 contribution of the capillary forces is negligible to powder cohesion in the conditions tested.

225 The particle morphology and surface of powders A, B and C can be observed in Fig. 1.

226 (Fig. 1)

227 In all samples one can notice particles with quite irregular shapes and rough surfaces,  
228 with asperities on them, characteristics which are similar to those reported in the literature for  
229 SCG powders (Ballesteros et al., 2014; Zarrinbakhsh, Wang, Rodriguez-Uribe, Misra, &  
230 Mohanty, 2016).

231

### 232 3.2 Effect of $d_s$ and particle size distribution on the flowability indexes

233 The measured values of HR are presented in Fig. 2 as a function of the mean diameter.

234 (Fig. 2)

235 The literature reports that flow indexes may be poorly reproducible because it is  
236 difficult to eliminate the past history of powders and to reproduce a macroscopically  
237 equivalent initial state in all the experiments (Castellanos, 2005). In the present study, the

238 standard deviations of HR (triplicate measurements) were in general lower than 5%,  
239 corroborating that the experimental procedures were effective to reduce variability.

240 In Fig. 2, the results for the base powders A, B and C are indicated by hollow circles,  
241 while the binary (AB, BC, and AC) and ternary mixtures (ABC) are indicated by solid  
242 symbols and hollow squares, respectively. The mixtures were obtained by combining the base  
243 powders, as described in Section 2.1, therefore some powders have similar mean sizes but  
244 different particle size distribution (PSD).

245 According to Fig. 2, the coarse powders A and B have similar HR ( $1.21\pm 0.03$  and  
246  $1.21\pm 0.05$ , respectively) and both are categorized as *free-flowing* materials. As for sample C,  
247 which is the fine powder, HR increases to  $1.50\pm 0.03$ , which characterizes a *very poor*  
248 flowability. The transition from a *passable* to a *poor* flowability ( $HR > 1.35$ ) is observed at a  
249 particle size of  $350\ \mu\text{m}$ . Some authors (Li et al., 2004) observed that HR can be a misleading  
250 flowability indicator for particles having high adhesiveness, or a very broad size distribution  
251 and irregular shapes and in tests with pharmaceutical powders they found discrepancies  
252 between the HR results in comparison to real flow situations. However, the cohesive behavior  
253 of powder C was clearly noticeable during the experiments and is corroborated by the AoR  
254 results, as will be discussed ahead.

255 The boundary size for a transition between cohesive and non-cohesive powders is not  
256 precisely defined in the literature, but it is agreed that the intensity of van der Waals forces is  
257 enhanced by the presence of particles of sizes less than  $100\ \mu\text{m}$ , owing to the increase in the  
258 specific superficial area and in the number of contacts between particles per unit of volume  
259 (Castellanos, 2005). To investigate the possible reasons for the poor flowability of powder C,  
260 its particle size distribution was measured using a laser diffraction technique, in Malvern  
261 Mastersizer (MAF5001). Triplicate measurements were performed and the data revealed that  
262 14% of the volume fraction in this sample was composed of particles with sizes less than  $100$

263  $\mu\text{m}$ . In spite of the small volume fraction, these fines correspond to about 93% of the specific  
264 superficial area in this sample. Therefore, the tendency towards a cohesive behavior of sample  
265 C is justified by the presence of these fine particles which were not detected in the sieving  
266 analysis, possibly because they agglomerate and are retained in the sieve with an aperture of  
267  $150 \mu\text{m}$ . Although the presence of fines is noticed in all the samples in the micrographs from  
268 Figure 1, the amount is greater in sample C and the impact of fines on powder cohesiveness is  
269 more significant in this sample owing to the substantial increase in the specific superficial  
270 area. The presence of agglomerates in sample C is corroborated by its high void fraction, as  
271 seen in Table 1.

272 The data in Fig. 2 shows that HR of the mixtures initially decreases steeply at a  
273 practically linear rate in the range of  $d_s$  from 225 to 550  $\mu\text{m}$ , and tend to a constant value  
274 close to 1.25 for particle sizes over 400  $\mu\text{m}$ . An exponential equation, represented by the solid  
275 line in Fig. 2, was fitted to the experimental data, with a correlation coefficient  $R^2=0.90$ . The  
276 equation is given by:

277

$$HR = 1 + e^{-0.003d_s} \quad (7)$$

278

279 The uniform distribution of the data around the fitting line in Fig. 2 suggests no  
280 evidence of an influence of the particle size distribution in HR.

281 The results for AoR are shown in Fig. 3. The wider dispersion in AoR measurements in  
282 comparison to HR can be attributed to the uncertainties that arise from the loose consolidation  
283 state of the conical heap formed in the AoR measurements, as the random arrangement of  
284 particles is hardly reproducible in different assays in spite of the attempts to standardize the  
285 experimental procedures. This problem is minimized in the HR measurements, as the initially  
286 loose-packed bed goes towards a quasi-consolidated state throughout tapping.

287 (Fig. 3)

288 In spite of the wide dispersion, the data show a consistent decrease in AoR as the  
289 particle size increases. Like the results for HR, the values of AoR for powders A and B are  
290 quite similar (respectively  $42\pm 1^\circ$  and  $42.3\pm 0.8^\circ$ ) and indicative of a *passable* flowability. It  
291 increased up to  $46.7\pm 0.3^\circ$  for powder C, characterizing a *very poor* flowability for this  
292 sample. The results for the powder mixtures are intermediate between powders A (the  
293 coarsest) and C (the finest). A polynomial equation was fitted to experimental data  
294 (represented by the solid line in Fig. 3), however, in view of the wide dispersion of data, the  
295 correlation coefficient was too low ( $R^2=0.43$ ). The equation is given by:

296

$$AoR = 91.99d_s^{-0.122} \quad (8)$$

297

298 As suggested by some authors (Geldart, Abdullah, Hassanpour, Nwoke, & Wouters,  
299 2006; Wouters & Geldart, 1996), the correlation could be improved by using the weighted  
300 AoR (defined as the ratio of AoR to the loose bulk density) to replace AoR.

301 It is worth noting in Fig. 3 that, even at similar mean particle sizes, the AoR of the  
302 binary mixtures made of A and C powders are always significantly higher than the ones  
303 measured for the mixtures of B and C, suggesting that the angle is sensitive to the relative size  
304 of powders in the mixture and also to particle size distribution.

305 The transition from a *passable* to a *poor* flowability ( $AoR > 45^\circ$ ) is observed for a  
306 particle size close to  $350 \mu\text{m}$ , which agrees with the results observed for HR. This result  
307 confirms that both indexes are good indicators of powder flowability and yields to consistent  
308 predictions, despite the mean size obtained from sieving is not representative of powders'  
309 PSD.

310 A comparison of HR and AoR of powders A and B to those of other biomasses with  
311 similar mean sizes (Lam & Sokhansanj, 2014) shows that SCG flowability indexes obtained  
312 in the present study are close to those reported for switchgrass, which is an ease flow biomass.  
313 It is worth noting that the flowability indexes of SCG were more sensitive to the variation of  
314 particle mean size than other regular biomasses. A decrease in  $d_s$  from 550 to 225  $\mu\text{m}$   
315 increased HR and AoR indexes of SCG by approximately 19.3 and 13.7%, respectively, while  
316 a similar variation increased HR and AoR of switchgrass, corn stover and wheat straw in only  
317 5%.

318 The correlation between the weighted AoR and HR can be observed in Fig. 4. The  
319 experimental data fitted an exponential equation with  $R^2=0.92$ . The equation is given by:

320

$$AoR/\rho_{lb} = 0.017e^{1.54HR} \quad (9)$$

321

322 The detail in Fig. 4 shows the data of the pure AoR as a function of HR, which clearly  
323 yields a significantly poorer fitting.

324 (Fig. 4)

325 As aforementioned, the AC binary mixtures (with coarser particles and wider size span),  
326 always presented worse flowability in comparison to the BC mixtures, which is evidence of  
327 the influence of the particle size distribution (PSD) on AoR. We observed that the loose bulk  
328 densities of AC mixtures were on average 5% higher than those of BC mixtures, which  
329 contributes to form taller heaps with higher AoR in AC mixtures.

330 Presenting the flowability of powders as a function of the mean diameter is a useful and  
331 common practice. However, this approach may lead to misleading conclusions as powders  
332 with different PSD and similar mean diameter can present distinct flow behavior. This is the  
333 case, for example, of the base powder B ( $d_s=400 \mu\text{m}$ ,  $AoR=42.3^\circ$  and  $HR=1.21$ ) and the

334 ternary mixture A<sub>60%</sub>B<sub>20%</sub>C<sub>20%</sub> ( $d_s=403 \mu\text{m}$ ,  $AoR=45.4^\circ$  and  $HR=1.30$ ). Based on these  
335 values, the flowability of the first powder is classified as *passable* and the second's as *poor*.

336 In the next section, the validity of the Modified Linear-Packing model (Yu & Standish,  
337 1991) to predict the loose and tapped packed-bed void fractions for the powder mixtures is  
338 evaluated.

339

#### 340 3.4 Modified Linear-Packing model to predict $\varepsilon_b$ of the SCGs mixtures

341 According to Figures 2 and 3, the flowability indexes of the binary and ternary mixtures  
342 are in between the values obtained for the largest (A) to the finest (C) powders. Therefore, the  
343 Modified Linear-Packing model (Yu, Bridgwater, & Burbidge, 1997) was applied to predict  
344 the void fraction of packed-beds of variable composition. The small-to-large ratios of the  
345 mixtures ( $R_{ij}$ ) were calculated from the mean sizes of A, B and C ( $d_{SA}>d_{SB}>d_{SC}$ ), and yielded  
346 the values  $R_{BA}=0.73$ ,  $R_{CB}=0.56$ , and  $R_{CA}=0.41$ . The interaction functions  $f(R_{ij})$  and  $g(R_{ij})$ ,  
347 that depend only on  $R_{ij}$ , were determined by Eqs. (10) and (11), recommended for spherical  
348 and non-spherical particles (Yu, Zou, & Standish, 1996).

349

$$350 \quad f(R_{ij}) = (1 - R_{ij})^{3.33} + 2.81 \cdot R_{ij} \cdot (1 - R_{ij})^{2.77} \quad (10)$$

$$351 \quad g(R_{ij}) = (1 - R_{ij})^{1.97} + 0.36 \cdot R_{ij} \cdot (1 - R_{ij})^{3.67} \quad (11)$$

352

353 The bed void fraction ( $\varepsilon$ ) was then calculated by Eq. (12), in which  $V$  is the specific  
354 volume of the powder mixture. In this study, the mixtures were produced by combining the  
355 three base powders, therefore variable  $V$  was obtained from Eqs. (13) to (16). The values of  
356  $V_A$ ,  $V_B$ , and  $V_C$  were calculated from Eq. (12) using the experimental values of loose or  
357 tapped void fractions (Table 1), according to the packing state to be determined.

358

$$\varepsilon = (V - 1)/V \quad (12)$$

$$V = \max\{V_A^T, V_B^T, V_C^T\} \quad (13)$$

$$V_A^T = V_A \cdot y_A + V_B \cdot (1 - f(R_{BA})) \cdot y_B + V_C \cdot [1 - f(R_{CA})] \cdot y_C \quad (14)$$

$$V_B^T = [V_A - (V_A - 1) \cdot g(R_{BA})] \cdot y_A + V_B \cdot y_B + V_C \cdot [1 - f(R_{CB})] \cdot y_C \quad (15)$$

$$V_C^T = [V_A - (V_A - 1) \cdot g(R_{CA})] \cdot y_A + [V_B - (V_B - 1) \cdot g(R_{CB})] \cdot y_B + V_C \cdot y_C \quad (16)$$

362

363 As shown in Fig. 5, the experimental void fractions are quite well predicted by the  
 364 model, for both loose and tapped conditions, with mean deviations less than 2%. The  
 365 experimental values of void fractions were estimated by Eq. 2 as the base samples present low  
 366 moisture content, as shown in Table 1.

367 (Fig. 5)

368 The loose and tapped predicted void fractions were used in Eq. (2) to calculate the bulk  
 369 densities of powders and HR was obtained for the binary and ternary mixtures. The predicted  
 370 (solid lines) and measured (symbols) values of HR are presented in Fig. 6 as a ternary  
 371 diagram, where each vertex corresponds to a pure base sample (A, B or C), the edges  
 372 correspond to the binary mixtures (AB, AC, and BC) with a variable mass fraction from 0 to  
 373 100%, and the interior corresponds to the ternary mixtures (ABC). For example, the point  
 374 marked with the star symbol on side line A-C of Fig. 6 shows that HR of a binary mixture  
 375 composed of 60% of A and 40% of C is equal to 1.33. For a ternary mixture with 60% of A,  
 376 20% of B and 20% of C (the star symbol inside the triangle), the value of HR is 1.26.  
 377 According to (Yu & Standish, 1991), when the ratio between particle sizes is higher than  
 378 0.154, as is the case of our mixtures, powder compaction occurs mainly by the occupation  
 379 mechanism, in which the voids are filled in throughout packing as the smaller particles  
 380 occupy the spaces opened by particle rearrangement. In a packing ruled by such a mechanism,  
 381 all the components in the mixture contribute to the compaction kinetic. In spite of that, HR of

382 the SCG mixtures was affected mostly by the mass fraction of the finer particles (powder C),  
383 as the smaller and more cohesive particles limit the packed-bed structure rearrangement in  
384 order to achieve a closer packing (Lam & Sokhansanj, 2014).

385 (Fig. 6)

386 Based on HR results shown in Fig. 6, it can be seen that powder flowability worsen as  
387 the mass fraction of powder C is raised. Therefore, powders with  $y_c=0$  have flowability  
388 categorized as *fair*; with  $y_c=0.2$  the flowability is *passable*; in the interval  $0.4 \leq y_c \leq 0.6$  it is  
389 *poor* and with  $y_c \geq 0.8$  it is *very poor*. For  $y_c > 0.5$  the HR increases linearly with the increase in  
390  $y_c$ , indicating that powder flowability is governed by the finest particles and the presence of  
391 coarse particles did not alter the flowability. As AoR is directly correlated to HR, as shown in  
392 Fig. 4 (or Eq. 9), the increase in the amount of powder C in a mixture also increases AoR.

393 It is worth noting that the ternary diagram proposed in Fig. 6 can be used to predict the  
394 HR for a sample sieved and cut into three fractions with mean sizes similar to the base  
395 samples A, B and C. New diagrams would need to be made for fractions with different sizes.  
396 Nevertheless, using a ternary diagram as a graphical representation of changes in HR could be  
397 a useful tool for a rapid and reliable estimation of HR and the method can be extended for  
398 other powders.

399 In the next section, the flowability indexes of SCG will be evaluated for moistures up to  
400 approximately 50% w.b., which cover the usual processing range (Silva et al., 1998).

401

### 402 3.6 Effect of moisture content on the flowability indexes

403 The values of HR and AoR data for powders under different moisture contents are  
404 presented in Table 2.

405 (Table 2)

406 As the moisture content of the base samples was raised up to 50% w.b., HR stays  
407 practically constant and AoR shows a discrete increase, although in most conditions the  
408 difference is within the range of experimental uncertainty. It is known from the literature that  
409 increasing moisture content is likely to affect the powder flow behavior and mechanical  
410 strength as the packing reaches the funicular state, at saturation levels between 25% and 50%,  
411 when saturated clusters and liquid bridges coexist (Althaus & Windhab, 2012). In the range of  
412 conditions tested, a slight deterioration in flowability was observed for powder C only at a  
413 saturation level close to 42%, corresponding to a moisture content of 59.2%. This indicates  
414 that the bridging forces were strong enough to significantly affect the flowability index in this  
415 condition. Therefore, for  $S < 42\%$  the samples may be in the pendular state, in which liquid  
416 bridges exist but do not play a significant role on the flow behavior. It is worth noting that the  
417 value of  $S = 42\%$  found in this study is between 25% and 50%, which is reported as a transition  
418 range from the pendular to the funicular states for real systems (Althaus & Windhab, 2012).

419 Since the dried and moist base samples presented similar HR up to 50% w.b., the  
420 diagram presented in Fig. 6 can be used for moist powders up to this range.

421

422 Based on the analysis of the previous sections, some alternatives to estimate HR and  
423 AoR of non-consolidated SCG powders are suggested:

- 424 • The Eq. (7), valid in the range  $225 < d_s < 550 \mu\text{m}$  can be used to estimate HR if the  
425 powder mean diameter is known. In addition, by measuring powder loose bulk  
426 density, Eq. (9) can provide estimates for AoR.
- 427 • If the powder size can be expressed as a composite of three mass fractions of  
428 different sizes, the ternary diagram presented in Fig. 6 can be used to estimate HR  
429 and AoR can be estimated from Eq. (9).

430 • Inversely, Eq. (9) can be used to obtain HR if powder AoR and loose bulk density  
431 are measured. Based on the measured values of HR, the ternary diagram can be used  
432 as a tool to evaluate the mass fraction of each base sample in the binary or ternary  
433 mixtures.

434 These procedures are based on quick and easy measurements and may be useful to help  
435 in process design and monitoring unit operations involved in renewable energy generation in  
436 soluble coffee facilities.

437

438

#### 439 **4. Conclusions**

440 This study presented experimental data of HR and AoR of non-consolidated SCG  
441 powder mixtures prepared from combinations of three base samples and with mean sizes from  
442 225 to 550  $\mu\text{m}$ . The measured indexes showed that a *poor* flowability is observed for SCG  
443 powders with  $d_s < 350 \mu\text{m}$  or for mixtures with a fraction greater than 40% in mass of fines.  
444 The AoR of SCG mixtures was influenced by the PSD, as samples with wider size span  
445 presented higher AoR. The HR of the mixtures increased significantly as the amount of the  
446 finest and more cohesive sample rose. The flowability behavior of moist SCGs was similar to  
447 that of dried powders for moisture contents up to 50% (w.b.). It was verified that SCGs  
448 present a higher energy density and worse flowability indexes when compared to other  
449 ordinary biomasses. Finally, some procedures to identify flowability indexes based on fitted  
450 correlations and on the Linear Mixture-Packing Model were proposed. These procedures may  
451 be useful to improve handling and monitoring operations in plants that use SCGs for  
452 thermochemical applications.

453

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457

## Nomenclature

AoR	Angle of repose, defined by Eq. (6) (°)
D	Mean diameter of the circumference of the conical powder bed (cm)
d	Diameter of the funnel discharge orifice (cm)
$d_s$	Mean diameter, defined by Eq. 1 ( $\mu\text{m}$ )
$d_{svi}$	Mean sieve diameter of a base sample, $i = A, B$ or $C$ ( $\mu\text{m}$ )
$f(R_{ij})$	Interaction function defined by Eq. (10) (-)
$g(R_{ij})$	Interaction function defined by Eq. (11) (-)
H	Height of the conical bed (cm)
HR	Hausner ratio, defined by Eq. (5) (-)
MC	Moisture content (%)
N	Number of taps (-)
p	Significance level (-)
$R_{ij}$	Ratio of small (i) to large (j) mean diameter (-)
S	Water saturation level, defined by Eq. (4) (-)
$y_i$	Mass fraction of a base sample in the mixture, $i = A, B$ or $C$ (-)
V	Specific volume, defined by Eq. (13) (-)
$V_i$	Specific volume of a base sample, $i = A, B$ or $C$ , determined by Eq. (12). (-)
$V_i^T$	Specific volume for $i = A, B$ or $C$ , defined by Eqs. (14) to (16)
$V_s$	Volume occupied by solids in the packed-bed ( $\text{m}^3$ )
$V_t$	Total volume of the dry tapped packed-bed ( $\text{m}^3$ )
$V_w$	Total volume of liquid water in the packed-bed ( $\text{m}^3$ )

## Greek letters

$\varepsilon$	Void fraction, defined by Eq. (2) (-)
$\varepsilon_{ib}$	Void fraction for loose bulk condition (-)
$\varepsilon_{tb}$	Void fraction for tapped bulk condition (-)

$\rho_{lb}$	Loose bulk density for N=0 (kg/m <sup>3</sup> )
$\rho_p$	Particle density (kg/m <sup>3</sup> )
$\rho_s$	Solid density (kg/m <sup>3</sup> )
$\rho_{tb}$	Tapped bulk density for N=1250 (kg/m <sup>3</sup> )

458

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543

Figure 1

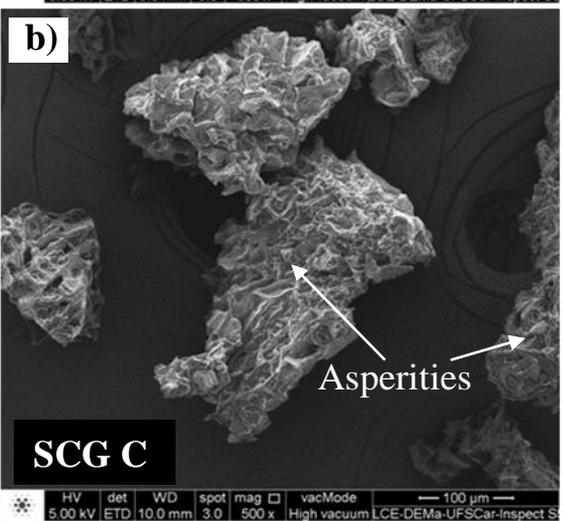
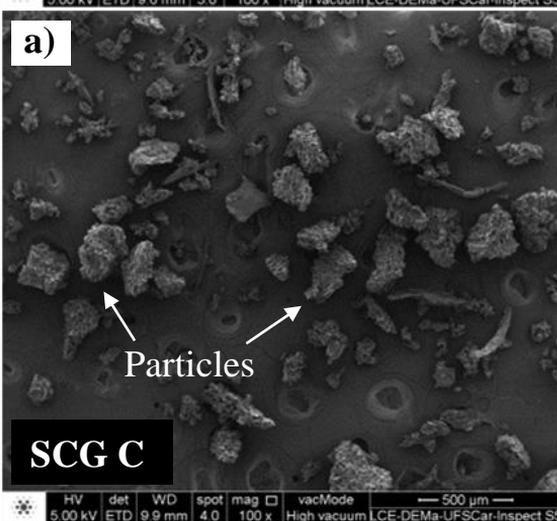
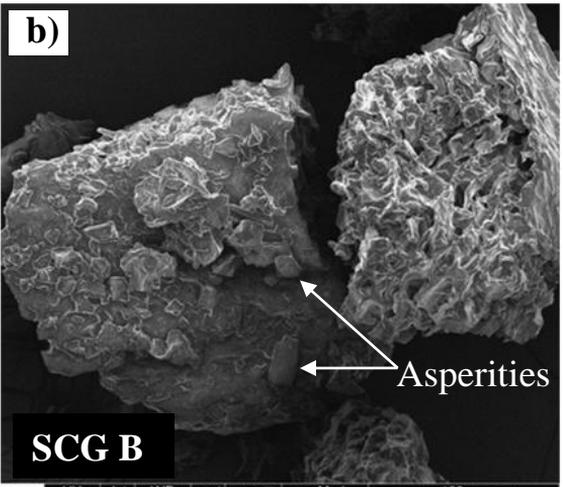
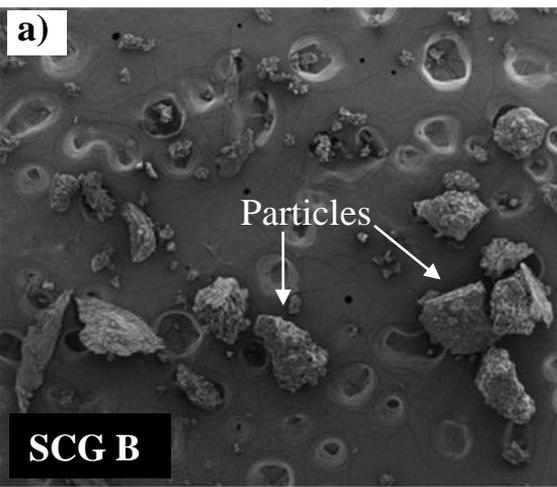
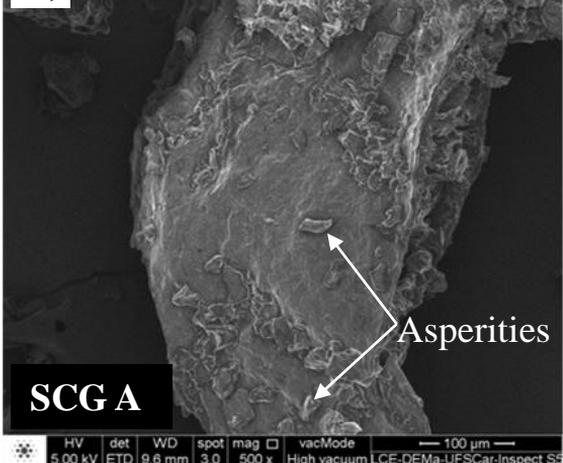
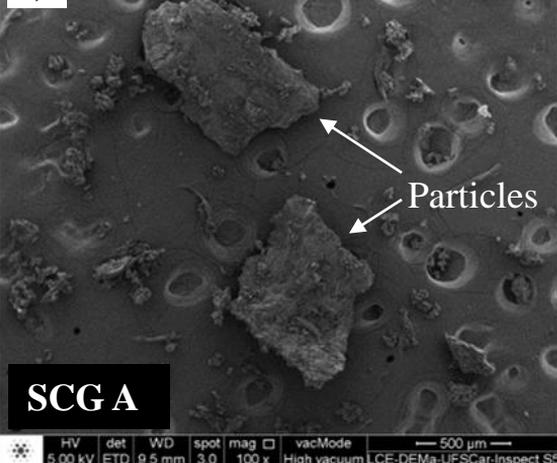
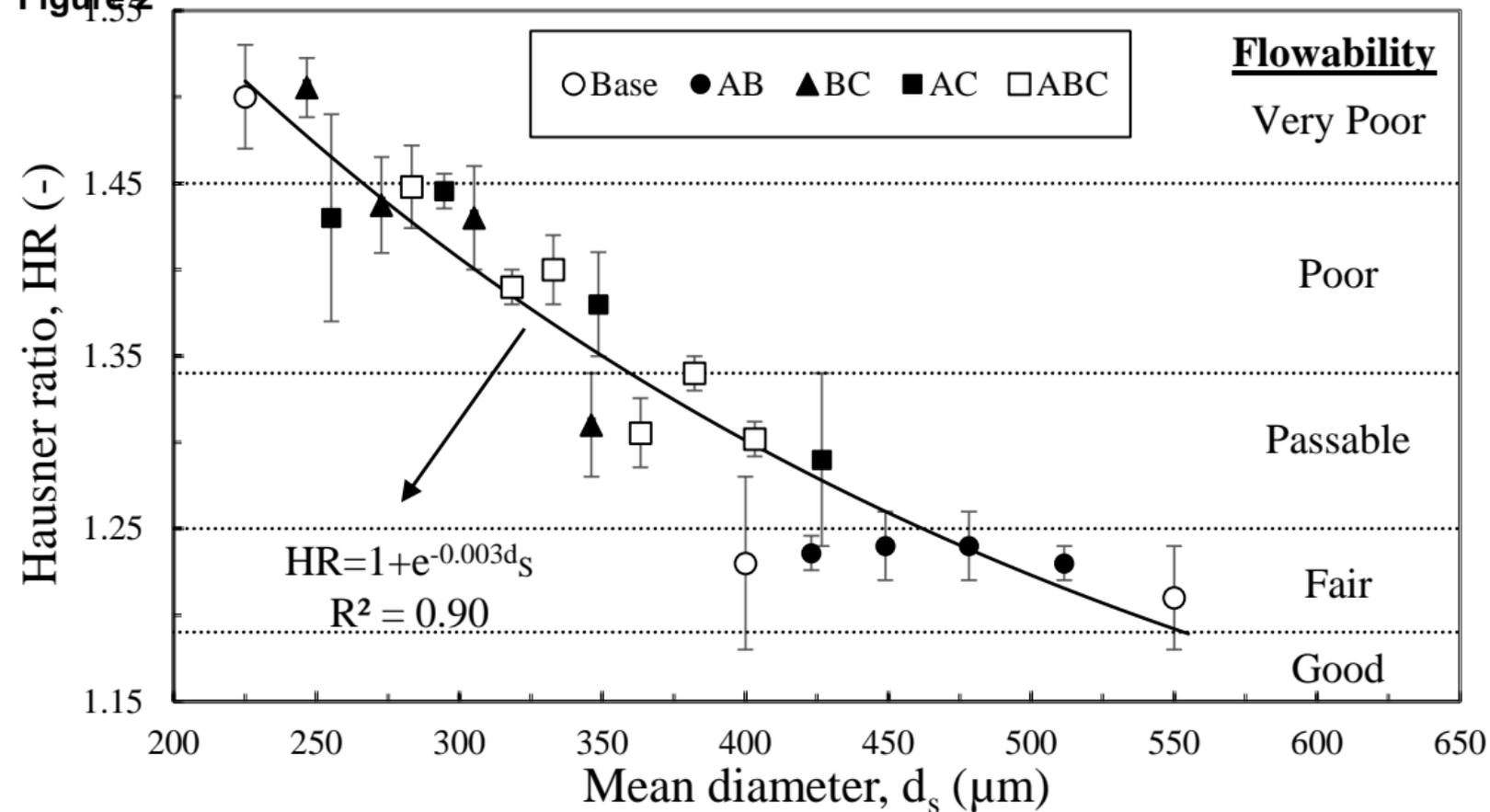
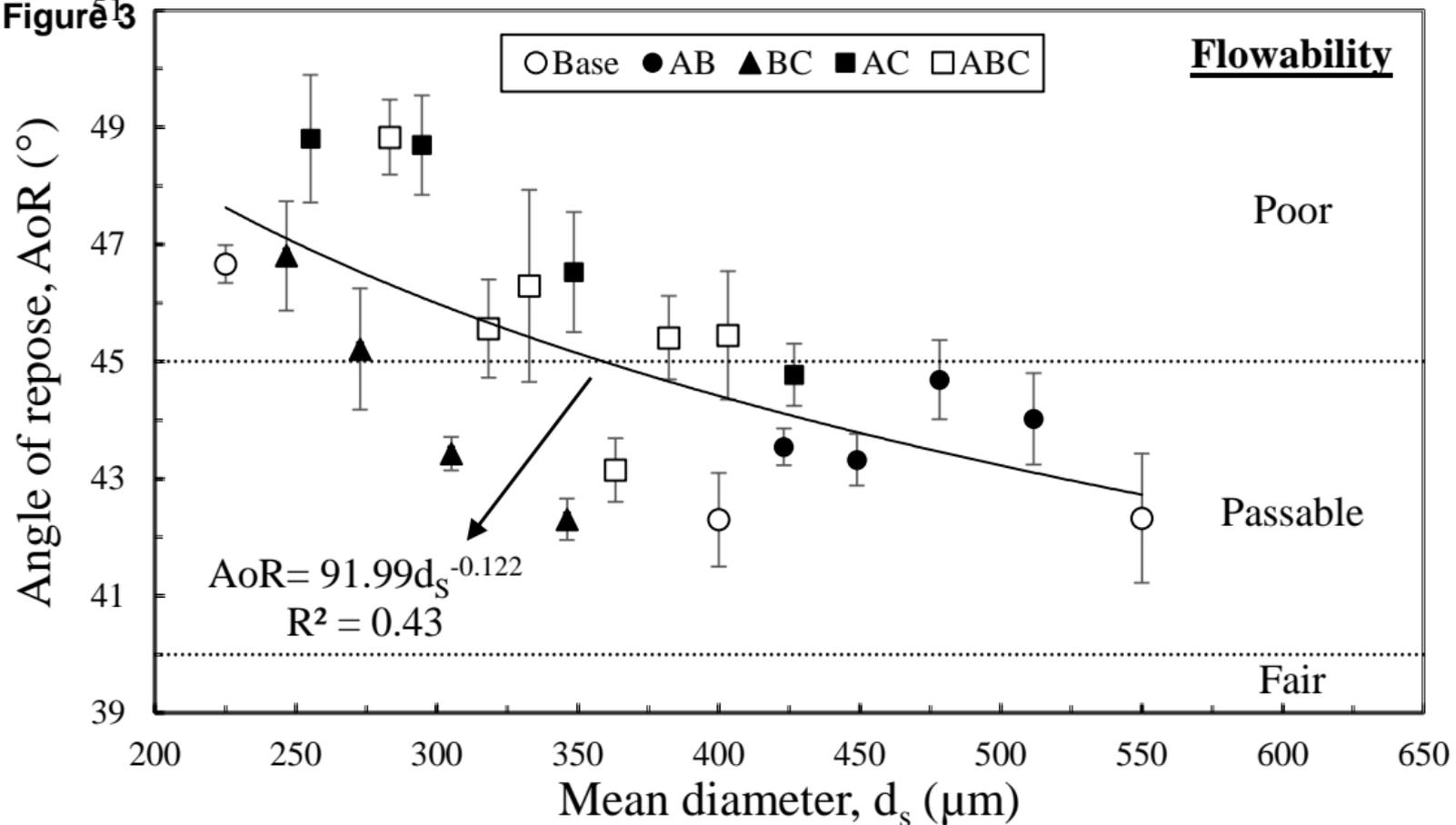


Fig. 1. Micrographs of A, B and C samples. (a) 100x and (b) 500x magnification.

**Figure 2**

**Fig. 2.** Hausner ratio as a function of mean diameter for the base samples, binary (AB, AC, and BC) and ternary mixtures (ABC) of SCG powders.

Figure 3



**Fig. 3.** Repose angle as a function of mean diameter for the base samples, binary (AB, AC, and BC) and ternary mixtures (ABC) of SCG.

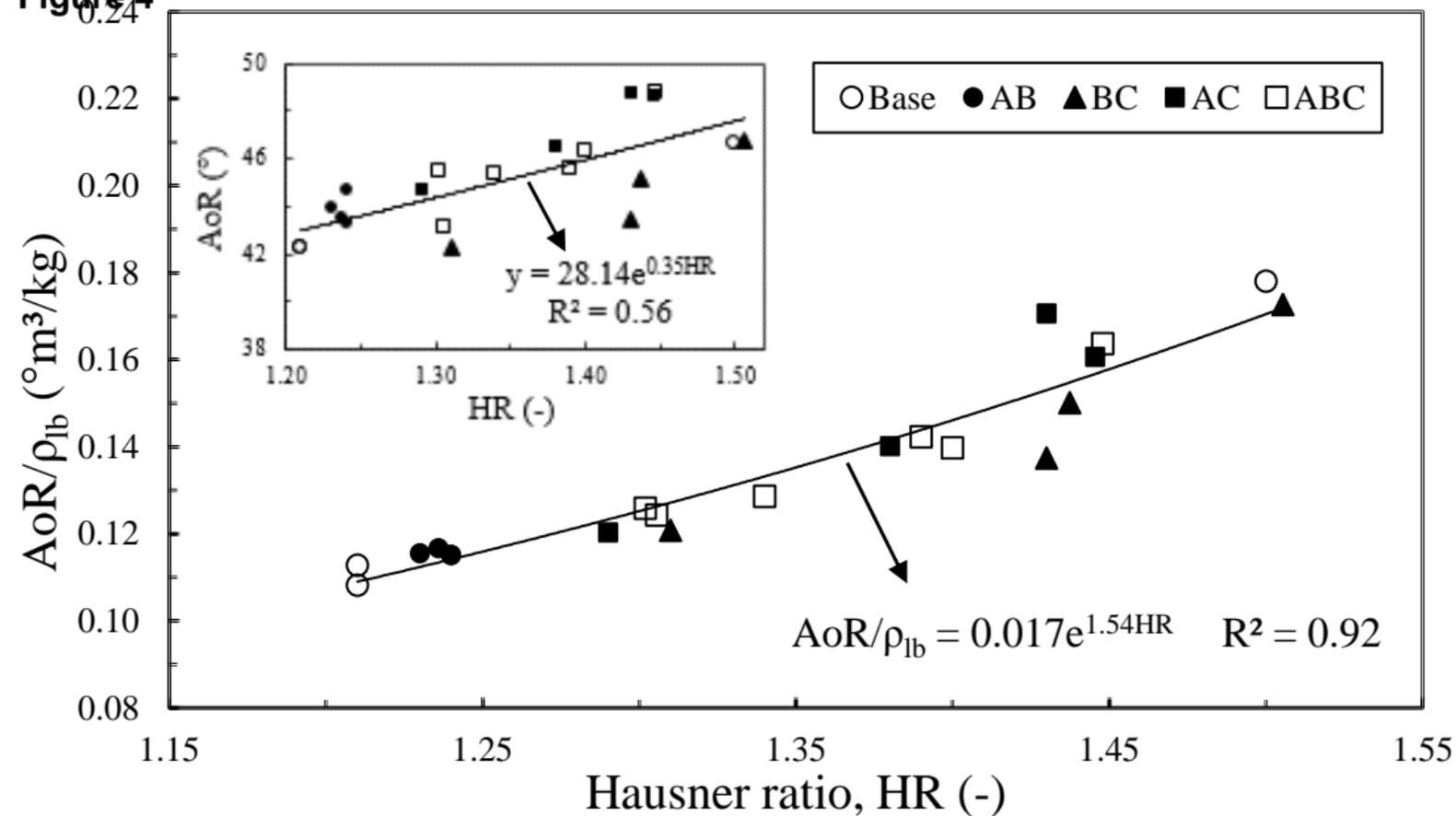
**Figure 4****Fig. 4.** Weighted AoR versus HR for SCG powders.

Fig. 5

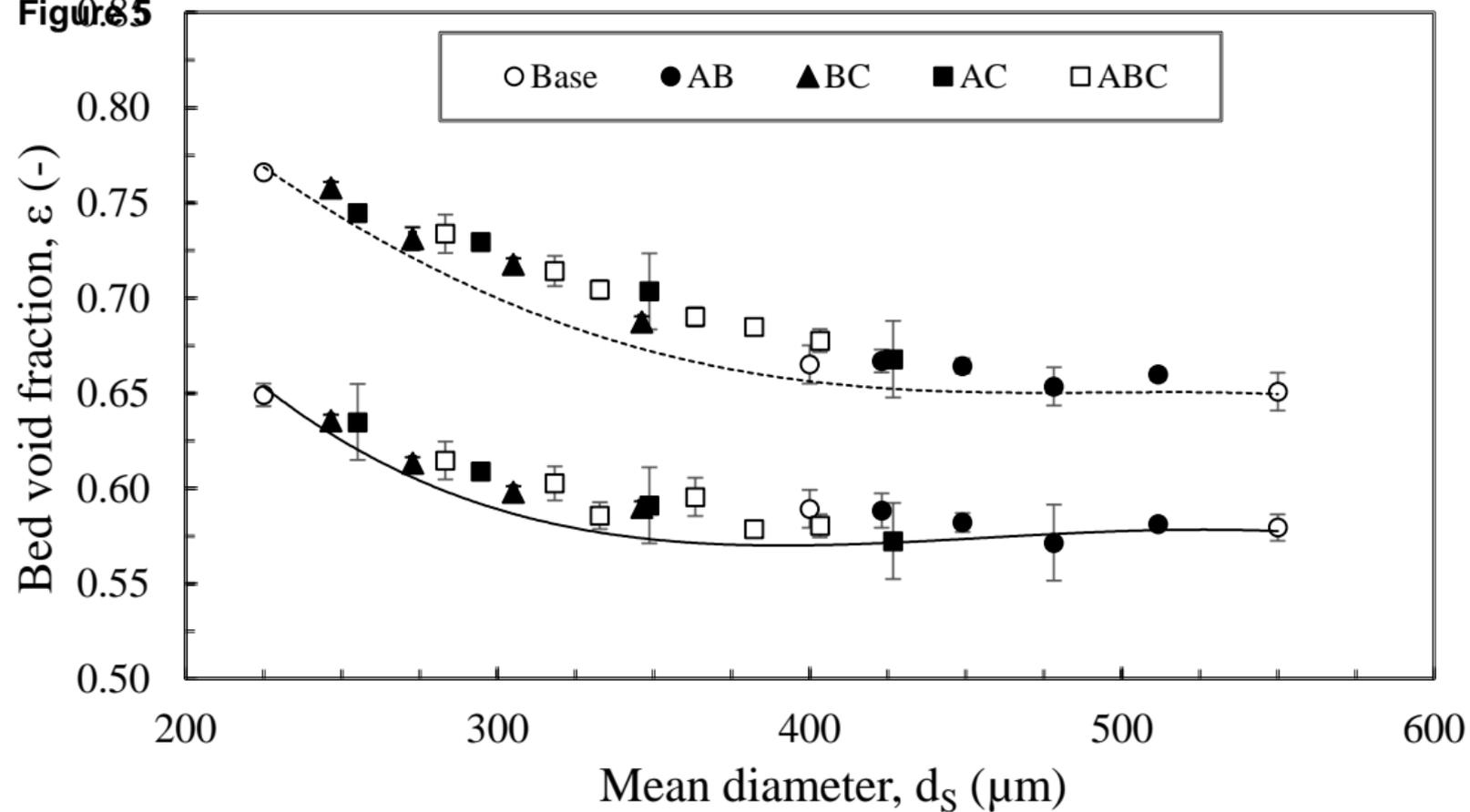


Fig. 5. Porosities as a function of mean diameter for the SCG base samples, binary and ternary mixtures: —, tapped and ---, loose porosities predicted by the Modified Linear-Packing model.

Figure 6

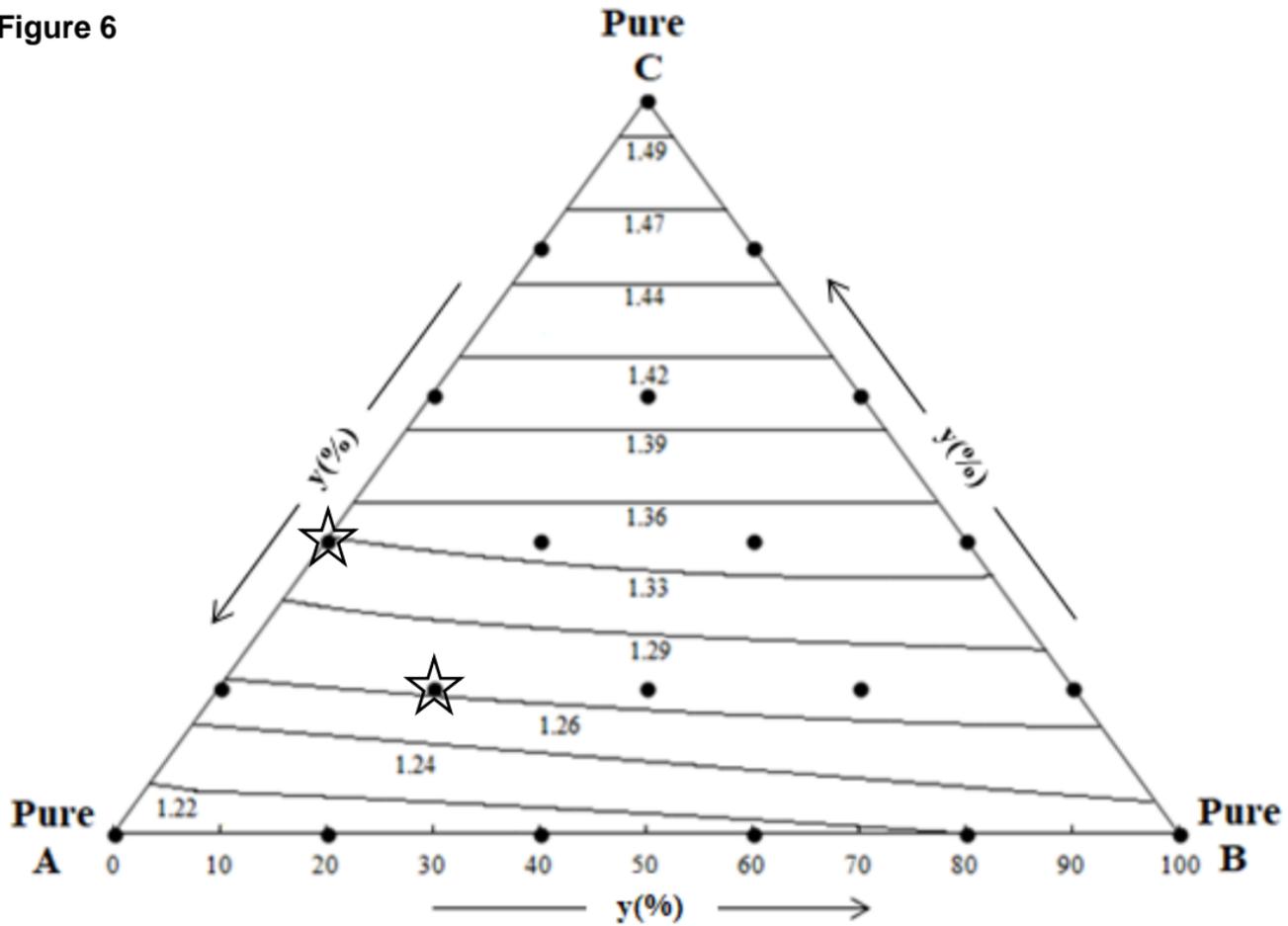


Fig. 6. HR values for SCG powders as a function of base samples composition (y): ●, HR measured points; —, predicted by the Modified Linear-Packing model.

**Table 1.** Sieve mean diameter, moisture content, particle density, loose and tapped bulk densities and porosities for the base powders.

SCG Powder	$d_{sv}$ ( $\mu\text{m}$ )	Moisture content (% w.b.)	$\rho_p$ ( $\text{kg/m}^3$ )	$\rho_{lb}$ ( $\text{kg/m}^3$ )	$\rho_{tb}$ ( $\text{kg/m}^3$ )	$\epsilon_{lb}$ (-)	$\epsilon_{tb}$ (-)
A	550	$5.4 \pm 0.1^a$	$1130 \pm 10^a$	$390 \pm 20^a$	$471 \pm 7^a$	$0.655 \pm 0.010^a$	$0.583 \pm 0.009^a$
B	400	$2.8 \pm 0.1^b$	$1120 \pm 20^a$	$380 \pm 20^a$	$460 \pm 10^a$	$0.660 \pm 0.010^a$	$0.590 \pm 0.010^a$
C	225	$3.2 \pm 0.1^c$	$1120 \pm 10^a$	$262 \pm 2^b$	$393 \pm 7^b$	$0.766 \pm 0.001^b$	$0.649 \pm 0.006^b$

Values with different letters in the same column are significantly different at a 0.05 significance level.

**Table 2.** Flowability of the base samples as a function of moisture content and water saturation level.

<b>SCG Sample</b>	<b>MC (% w.b.)</b>	<b>S (%)</b>	<b>HR (-)</b>	<b>AoR (°)</b>
<b>A</b>	5.4 ± 0.1	3.8 ± 0.1	1.21 ± 0.03 <sup>ab</sup>	42.0 ± 1.0 <sup>ab</sup>
	21.1 ± 0.5	13.3 ± 0.3	1.19 ± 0.01 <sup>ab</sup>	43.0 ± 1.0 <sup>ab</sup>
	30.4 ± 0.2	17.3 ± 0.1	1.17 ± 0.01 <sup>ab</sup>	43.2 ± 0.7 <sup>ab</sup>
	37.9 ± 0.1	20.8 ± 0.1	1.17 ± 0.01 <sup>a</sup>	43.2 ± 0.6 <sup>ab</sup>
	47.0 ± 1.0	27.3 ± 1.0	1.21 ± 0.02 <sup>b</sup>	44.1 ± 0.4 <sup>b</sup>
<b>B</b>	2.8 ± 0.1	2.1 ± 0.1	1.21 ± 0.05 <sup>a</sup>	42.3 ± 0.8 <sup>a</sup>
	19.4 ± 0.9	12.5 ± 0.3	1.23 ± 0.02 <sup>a</sup>	45.8 ± 0.2 <sup>b</sup>
	28.6 ± 0.7	16.1 ± 0.5	1.21 ± 0.01 <sup>a</sup>	45.0 ± 1.0 <sup>b</sup>
	36.8 ± 0.9	20.0 ± 0.8	1.25 ± 0.02 <sup>a</sup>	44.8 ± 0.8 <sup>b</sup>
	44.5 ± 0.5	26.3 ± 0.3	1.30 ± 0.07 <sup>a</sup>	46.1 ± 0.5 <sup>b</sup>
<b>C</b>	3.2 ± 0.1	1.8 ± 0.1	1.50 ± 0.03 <sup>ab</sup>	46.7 ± 0.3 <sup>a</sup>
	29.6 ± 0.3	14.5 ± 0.4	1.45 ± 0.04 <sup>ab</sup>	48.7 ± 0.9 <sup>b</sup>
	42.0 ± 0.1	21.6 ± 0.2	1.44 ± 0.02 <sup>a</sup>	50.9 ± 0.2 <sup>c</sup>
	52.4 ± 0.4	31.1 ± 0.6	1.53 ± 0.02 <sup>b</sup>	52.0 ± 0.5 <sup>c</sup>
	59.2 ± 0.1	42.0 ± 0.2	1.62 ± 0.05 <sup>c</sup>	54.9 ± 0.3 <sup>d</sup>

Values with different letters in the same column are significantly different at a 0.05 significance level.