



**Greenwich Academic Literature Archive (GALA)**  
– the University of Greenwich open access repository  
<http://gala.gre.ac.uk>

---

*Citation for published version:*

Rudrangi, Shashi Ravi Suman, Kaialy, Waseem, Ghori, Muhammad U, Trivedi, Vivek, Alexander, Bruce and Snowden, Martin (2016) Solid-state flurbiprofen and methyl- $\beta$ -cyclodextrin inclusion complexes prepared using a single-step, organic solvent-free supercritical fluid process. *European Journal of Pharmaceutics and Biopharmaceutics*. ISSN 0939-6411 (doi:10.1016/j.ejpb.2016.04.024)

*Publisher's version available at:*

<http://www.sciencedirect.com/science/article/pii/S0939641116301655>

---

**Please note that where the full text version provided on GALA is not the final published version, the version made available will be the most up-to-date full-text (post-print) version as provided by the author(s). Where possible, or if citing, it is recommended that the publisher's (definitive) version be consulted to ensure any subsequent changes to the text are noted.**

*Citation for this version held on GALA:*

Rudrangi, Shashi Ravi Suman, Kaialy, Waseem, Ghori, Muhammad U, Trivedi, Vivek, Alexander, Bruce and Snowden, Martin (2016) Solid-state flurbiprofen and methyl- $\beta$ -cyclodextrin inclusion complexes prepared using a single-step, organic solvent-free supercritical fluid process. London: Greenwich Academic Literature Archive.  
Available at: <http://gala.gre.ac.uk/15496/>

---

Contact: [gala@gre.ac.uk](mailto:gala@gre.ac.uk)

**Solid-state flurbiprofen and methyl- $\beta$ -cyclodextrin inclusion complexes prepared using a single-step, organic solvent-free supercritical fluid process.**

Shashi Ravi Suman Rudrangi<sup>a, d\*</sup>, Waseem Kaialy<sup>b</sup>, Muhammad U. Ghori<sup>c</sup>, Vivek Trivedi<sup>a</sup>, Martin J. Snowden<sup>a</sup>, Bruce David Alexander<sup>a</sup>

<sup>a</sup>Department of Pharmaceutical, Chemical and Environmental Sciences, Faculty of Engineering and Science, University of Greenwich at Medway, Central Avenue, Chatham Maritime, Kent ME4 4TB, United Kingdom.

<sup>b</sup>School of Pharmacy, Faculty of Science and Engineering, University of Wolverhampton, Wolverhampton, West Midlands WV1 1LY, United Kingdom.

<sup>c</sup>Department of Pharmacy, University of Huddersfield, Queensgate, Huddersfield HD1 3DH, United Kingdom.

<sup>d</sup>Research and Development Department, Bristol Laboratories Limited, Laporte Way, Luton, Bedfordshire LU4 8WL, United Kingdom

\*Corresponding author. Tel: +44 (0) 7921530229, Fax: +44 (0) 2083319805.

Email address: suman\_rudrangijips@yahoo.com; Shashi@bristol-labs.co.uk

**ABBREVIATIONS**

DE, dissolution efficiency; DP, percent drug dissolved; DSC, differential scanning calorimetry; Me- $\beta$ -CD, methyl- $\beta$ -cyclodextrin; SC-CO<sub>2</sub>, supercritical carbon dioxide; SEM, rpm, revolutions per minute; Scanning electron microscopy; XRPD: X-ray powder diffraction.

## ABSTRACT

The aim of this study was to enhance the apparent solubility and dissolution properties of flurbiprofen through inclusion complexation with cyclodextrins. Especially, the efficacy of supercritical fluid technology as a preparative technique for the preparation of flurbiprofen-methyl- $\beta$ -cyclodextrin inclusion complexes was evaluated. The complexes were prepared by supercritical carbon dioxide processing and were evaluated by solubility, differential scanning calorimetry, X-ray powder diffraction, scanning electron microscopy, practical yield, drug content estimation and in vitro dissolution studies. Computational molecular docking studies were conducted to study the possibility of molecular arrangement of inclusion complexes between flurbiprofen and methyl- $\beta$ -cyclodextrin. The studies support the formation of stable molecular inclusion complexes between the drug and cyclodextrin in a 1:1 stoichiometry. In vitro dissolution studies showed that the dissolution properties of flurbiprofen were significantly enhanced by the binary mixtures prepared by supercritical carbon dioxide processing. The amount of flurbiprofen dissolved into solution alone was very low with  $1.11 \pm 0.09\%$  dissolving at the end of 60 min, while the binary mixtures processed by supercritical carbon dioxide at 45 °C and 200 bar released  $99.39 \pm 2.34\%$  of the drug at the end of 30 min. All the binary mixtures processed by supercritical carbon dioxide at 45 °C exhibited a drug release of more than 80% within the first 10 min irrespective of the pressure employed. The study demonstrated the single step, organic solvent-free supercritical carbon dioxide process as a promising approach for the preparation of inclusion complexes between flurbiprofen and methyl- $\beta$ -cyclodextrin in solid-state.

**Keywords:** Dissolution rate; Flurbiprofen; Inclusion complexes; Methyl- $\beta$ -cyclodextrin; Solubility; Supercritical fluid technology.

## 1. Introduction

The water solubility of a drug is a key indicator for the solubility of drug molecules in the intestinal fluids and its contribution to bioavailability. The success of a pharmaceutical formulation relies on how effectively it makes the drug molecules available at the site of absorption. However, poor water solubility and the resulting poor oral bioavailability of most drugs available on the market has long been a problem to pharmaceutical scientists, as the dissolution is quite often the rate-limiting process in the absorption of a drug (Charman and Stella, 1991; Takagi et al., 2006). The ability to increase the water solubility of a drug is thus very important to enhancing its therapeutic efficacy. Several strategies to improve the water solubility and dissolution rate of poorly soluble drugs have been reported (Rudrangi et al., 2011; Bontha et al., 2012; Weerapol et al., 2015; Kaialy and Al Shafiee, 2015; Aroso et al., 2016; Jiang et al., 2016), among which is the inclusion complexation with cyclodextrins (Bandi et al., 2004; Cirri et al., 2005; Vega et al., 2013; Rudrangi et al., 2015a; Rudrangi et al., 2015b).

Cyclodextrins are a family of non-reducing cyclic oligosaccharides composed of ( $\alpha$ -1,4)-linked  $\alpha$ -D-glucopyranose units. They possess a hydrophobic cavity and a hydrophilic exterior. The hydrophobic cavity can accommodate a poorly water-soluble drug, whilst the hydrophilic exterior facilitates high water solubility.

Rudrangi and co-workers carried out the phase solubility studies of indomethacin and olanzapine formulated with Me- $\beta$ -CD in phosphate buffer medium (pH 7.4) and de-ionised water (pH 7.1) at 37 °C, respectively. Rudrangi et al. (2015a) reported an A<sub>N</sub>-subtype phase solubility diagram with a stability constant of  $K_{1:1} = 167 \text{ M}^{-1}$  for the indomethacin-Me- $\beta$ -CD complexes while Rudrangi et al. (2015b) reported an A<sub>L</sub>-subtype phase solubility diagram with a stability constant of  $K_{1:1} = 304 \text{ M}^{-1}$  for the olanzapine-Me- $\beta$ -CD complexes.

Computational molecular docking studies of indomethacin and olanzapine with Me- $\beta$ -CD were carried out by Rudrangi and co-workers using the Glide application. The binding affinity (GLIDE energy), van der Waals energy and docking score for inclusion of indomethacin in Me- $\beta$ -CD were 27.880 kcal mol<sup>-1</sup>, -28.941 kcal mol<sup>-1</sup> and -4.882 kcal mol<sup>-1</sup>, respectively (Rudrangi et al., 2015a), while the binding affinity, Van der Waals energy and docking score for inclusion of olanzapine in Me- $\beta$ -CD were -24.13 kcal mol<sup>-1</sup>, -21.57 kcal mol<sup>-1</sup> and -3.09 kcal mol<sup>-1</sup>, respectively (Rudrangi et al., 2015b).

Flurbiprofen [2-(2-fluoro-4-biphenyl)propionic acid] is a potent chiral non-steroidal anti-inflammatory drug approved by the Food and Drug Administration for the treatment of rheumatoid arthritis and osteoarthritis (Davies, 1995). It also has profound analgesic effects and is indicated in the treatment of gingivitis and alveolar bone resorption in periodontitis (Heasman, P.A. et al., 1990). However, it has poor water solubility (5-13  $\mu$ g mL<sup>-1</sup>) and dissolution rate (Anderson and Conradi, 1985), which limits both its therapeutic application and efficacy. Hence, there is a need to enhance the water solubility and dissolution rate of flurbiprofen.

Several formulation strategies have been reported to improve the solubility and dissolution rate of flurbiprofen, such as dry elixir (Kim and Yoon, 1995), solid dispersion (Habib et al., 1998; Oh et al., 2011), salt formation (Gupta et al., 1997), microemulsion (Park et al., 1997; Ambade et al., 2008), inclusion complex with cyclodextrins (Tokumura et al., 2009; Li et al., 2010) and cycloamyloses (Baek et al., 2011).

Flurbiprofen has an affinity for different cyclodextrins forming inclusion complexes. The favourable effect of natural  $\beta$ -cyclodextrin (Muraoka et al., 2004; Cirri et al., 2005; Tokumura et al., 2009; Li et al., 2010; Baek et al., 2011) and its derivatives, such as hydroxypropyl- $\beta$ -cyclodextrin (Govindarajan and Nagarsenker, 2005; Vega et al., 2013),

hydroxyethyl- $\beta$ -cyclodextrin and methyl- $\beta$ -cyclodextrin (Cirri et al., 2005) on its pharmaceutical properties has been previously reported. For example, Cirri et al. (2005) reported that flurbiprofen and methyl- $\beta$ -cyclodextrin (Me- $\beta$ -CD) complexes prepared in the 1:1 molar ratio using kneading and co-evaporation techniques exhibit higher solubility and dissolution profiles than flurbiprofen alone or in a state of a physical mixture. In spite of their success, the complexes prepared by the co-evaporation and kneading methods required an organic solvent. Organic solvents are potentially hazardous and their presence in the final product (complexes) is highly undesirable and often requires energy-intensive drying steps involving a great deal of time to remove the residual solvent (Al-Marzouqi et al., 2007; Al-Marzouqi et al., 2009). Hence, there is a need to avoid the use of organic solvents in the preparation of drug-cyclodextrin complexes.

Pharmaceutical scientists relentlessly try to avoid the usage of toxic organic solvents and exchange those to more environmentally friendly alternatives with similar properties. One of those effective alternatives is the use of supercritical fluids. A supercritical fluid is a substance that exists above its critical point, where the phase boundaries diminish. Carbon dioxide is the most commonly employed supercritical fluid because of its low critical parameters. Supercritical carbon dioxide (SC-CO<sub>2</sub>) has provided an excellent green alternative to harmful organic solvents and the use of SC-CO<sub>2</sub> processing technique in the preparation of drug-cyclodextrin complexes has been reported (York, 1999; Kompella and Koushik, 2001; Sunkara and Kompella, 2002; Türk et al., 2007; Lee et al., 2008; Rudrangi et al., 2015a; Rudrangi et al., 2015b).

The application of supercritical carbon dioxide processing in the preparation of flurbiprofen-Me- $\beta$ -CD complexes has not yet been investigated. Therefore, the aim of this study was to evaluate supercritical fluid technology as a preparative technique for the preparation of

flurbiprofen-Me- $\beta$ -CD inclusion complexes in solid-state. Inclusion complexes were prepared by physical mixing and SC-CO<sub>2</sub> processing at various working (temperature and pressure) conditions. The prepared complexes were then characterized by solubility studies, differential scanning calorimetry (DSC), X-ray powder diffraction (XRD), scanning electron microscopy (SEM), practical yield, drug content evaluation and dissolution studies.

## 2. Materials and methods

### 2.1. Materials

Flurbiprofen ( $\geq 98.5\%$ , molecular weight: 244.26 g/mol) and Me- $\beta$ -CD (average molecular weight: 1310 g/mol, extent of labeling: 1.6-2.0 mol CH<sub>3</sub> per unit anhydroglucose) were purchased from Sigma-Aldrich (Gillingham, Dorset, UK). Ethanol was obtained from Fisher Scientific (Loughborough, UK). Carbon dioxide (99.9%) was obtained from BOC Ltd (Guildford, Surrey, UK). All reagents were used as received.

### 2.2. Preparation of binary mixtures of flurbiprofen with Me- $\beta$ -CD

All binary mixtures were prepared in a 1:1 molar ratio of flurbiprofen to Me- $\beta$ -CD. The processed samples were stored in desiccators until further use.

#### 2.2.1. Physical mixing

Required quantities of flurbiprofen and Me- $\beta$ -CD were accurately weighed and tumble-mixed at 100 rpm for 15 min using a Turbula<sup>®</sup> T2F mixer (Willy A. Bachofen AG – Maschinenfabrik, Muttenz, Switzerland).

#### 2.2.2. Supercritical carbon dioxide process

The SC-CO<sub>2</sub> process was carried out in the static mode to achieve the inclusion of flurbiprofen in the Me- $\beta$ -CD. The complexes were prepared using an extraction apparatus supplied by Thar Process Inc., USA as described in detail elsewhere (Rudrangi et al., 2015b).

The physical mixtures of flurbiprofen and Me- $\beta$ -CD were placed in a sample cell. Carbon dioxide was pumped from a cylinder via a cooling unit into the sample cell. The physical mixtures were processed at six different working conditions [35 °C/100 bar, 35 °C/150 bar, 35 °C/200 bar, 45 °C/100 bar, 45 °C/150 bar and 45 °C/200 bar] in order to study the influence of temperature and pressure on the inclusion complex formation.

### 2.3. Analysis of the prepared binary mixtures

Having prepared these model complexes by supercritical carbon dioxide processing, the next aim is to compare the physical properties with those of complexes prepared using physical mixing.

#### 2.3.1. Solubility studies

Saturation solubility of flurbiprofen was measured in triplicate by adding excess amounts of the drug to 10 mL of deionised water in sealed glass containers. The solutions were agitated for 72 hours at 25 °C. The solutions were then filtered (0.45  $\mu$ m filter pore size) and assayed for drug concentration by ultra-violet spectroscopy (Cary 100 UV-vis, Agilent Technologies, USA) after dilution.

Phase solubility diagram was obtained according to the method reported by Higuchi and Connors (1965). Samples were prepared by adding excess flurbiprofen (in an amount above its solubility), to a 10 mL deionised water containing successively increasing concentrations (0, 2.5, 5, 10, 12.5, 15, 20 and 25  $\times 10^{-3}$  M) of Me- $\beta$ -CD, in sealed glass containers. The solutions were agitated (200 rpm) at 25 °C for 72 hours. Following equilibrium, the solutions



were then filtered and assayed for drug concentration by UV spectroscopy as described above. Phase-solubility diagram was represented as the concentration of total dissolved flurbiprofen against the concentration of Me- $\beta$ -CD. The binding constant ( $K_{1:1}$ ) for flurbiprofen-Me- $\beta$ -CD complexes was calculated from the slope of the curve.

### 2.3.2. Differential scanning calorimetry (DSC) analysis

DSC analysis of flurbiprofen, Me- $\beta$ -CD and the flurbiprofen-Me- $\beta$ -CD complex systems was carried out using a FP-90 central processor (Mettler-Toledo, LLC, UK). The instrument was calibrated using indium as the reference material. For each sample, about 5 mg of each sample was accurately weighed, hermetically sealed in an aluminium pan, and then heated from 50 to 150 °C at a heating rate of 10 °C min<sup>-1</sup>. The control study was carried out to investigate the effect of SC-CO<sub>2</sub> processing on flurbiprofen alone.

### 2.3.3. X-ray powder diffraction (XRPD) analysis

X-ray powder diffractograms of flurbiprofen, Me- $\beta$ -CD and the flurbiprofen-Me- $\beta$ -CD complex systems were obtained on a D8 Advance X-ray Diffractometer (Bruker, Germany) in theta-theta Bragg-Brentano geometry using reflection mode. Data were collected between 2-40° 2 $\theta$ , with a step size of 0.006° and a time of 0.5 s per step using Cu K $\alpha$  radiation. A control study was carried out to investigate the effect of SC-CO<sub>2</sub> processing on flurbiprofen alone.

The degree of crystallinity (% crystallinity) of the flurbiprofen-Me- $\beta$ -CD complex systems was determined using the amorphous subtraction method as reported previously ([Rudrangi et al., 2015a](#)).

### 2.3.4. Practical yield

The efficiency of a processing method used for the preparation of inclusion complexes can be measured by determining the practical yield of the binary systems. The percentage practical yield of the binary systems was determined using the following equation:

$$\text{Practical yield (\%)} = \left[ \frac{\text{Practical weight of the binary mixture}}{\text{Theoretical weight (Drug + Cyclodextrin)}} \right] \times 100$$

### 2.3.5. Evaluation of the flurbiprofen content

Flurbiprofen-Me- $\beta$ -CD binary system, equivalent to 10 mg of the drug, was accurately weighed and dispersed in ethanol (10 mL), in which both the drug and cyclodextrin are soluble. The mixture was agitated (100 rpm) at  $25 \pm 0.5$  °C for 30 min and the resultant solution was filtered (0.45  $\mu$ m filter pore size) and assayed for the concentration of total flurbiprofen by UV spectroscopy after dilution. Determination of the flurbiprofen content in the complexes allows the verification that there is no loss of flurbiprofen during the (complex) preparation process.

### 2.3.6. In vitro dissolution studies

In vitro dissolution studies of flurbiprofen from all drug-carrier binary systems, and for flurbiprofen alone, were conducted in triplicate using USP Type II paddle method (Hanson G2 Vision<sup>®</sup> Classic 6, Chatsworth, CA) with deionised water as the dissolution medium. The samples, corresponding to 1000 mg of flurbiprofen were dispersed into 900 mL of the dissolution media maintained at  $25 \pm 0.5$  °C and stirred at 100 rpm. Aliquots (5 mL) were withdrawn and filtered (0.45  $\mu$ m filter pore size) at the specified time intervals (2, 5, 10, 20, 30 and 60 min) and the drug concentration was determined by UV spectroscopy at a  $\lambda_{\text{max}}$  of 246 nm. The same volume of fresh medium was added to the vessel and the correction for the cumulative dilution was calculated. The parameters used to characterize the dissolution curves were the percentage of flurbiprofen dissolved at 30 min, and the dissolution efficiency

at 30 and 60 min. The dissolution efficiency was calculated according to Khan (1975).  $DE_{30\text{min}}$  was calculated from the area under the dissolution curve at 30 min and expressed as a percentage of the area of the rectangle described by 100% dissolution in the same time of the total amount added.

### 2.3.7. Molecular docking studies

Molecular docking studies were conducted to study the possibility of molecular arrangement of inclusion complexes between flurbiprofen and Me- $\beta$ -CD. Molecular docking studies of flurbiprofen into Me- $\beta$ -CD were conducted using Glide (grid-based ligand docking) application implemented in the Maestro 9.3 software package (Schrodinger, LLC, New York, USA, 2012) as reported earlier (Rudrangi et al., 2015b). Docked flurbiprofen and Me- $\beta$ -CD complexes were visualised and molecular surface complex pictures were generated using Maestro.

## 3. Results and discussions

### 3.1. Solubility studies

The amount of flurbiprofen dissolved at 25 °C was  $12.29 \pm 1.05 \mu\text{g mL}^{-1}$  and  $30.68 \pm 0.5 \mu\text{g mL}^{-1}$  after 1 h and 24 h, respectively. No further improvement was observed in drug solubility after 24 h.

Cirri et al. (2005), Li et al. (2010) and Baek et al. (2011) reported a drug solubility of  $40 \mu\text{g mL}^{-1}$ ,  $10.45 \pm 3.22 \mu\text{g mL}^{-1}$  and  $5.12 \pm 1.22 \mu\text{g mL}^{-1}$  in water at 25 °C after 24 h, 120 h and 168 h, respectively. Herzfeldt and Kummel (1983) reported a drug solubility of  $8 \mu\text{g mL}^{-1}$  and  $31.2 \mu\text{g mL}^{-1}$  in pH 1.2 and 7.5 buffered solutions at 25 °C after 24 h, respectively. Varma and Pandi (2005) reported a drug solubility of  $48.2 \mu\text{g mL}^{-1}$  in water at 27 °C after 72 h.

The solubility of flurbiprofen increased in a linear fashion with the concentration of Me- $\beta$ -CD (Fig. 1), forming A<sub>L</sub>-subtype complexes (presumably 1:1 stoichiometry). The binding constant of the complexes ( $K_{1:1} = 16856 \pm 1244 \text{ M}^{-1}$ ) was calculated from the slope of the curve. Similar phase solubility profile was reported for flurbiprofen-Me- $\beta$ -CD complexes in water by Cirri et al. (2005). The authors reported the stability constant values of  $K_{1:1} = 11570 \text{ M}^{-1}$  and  $9660 \text{ M}^{-1}$  in unbuffered water (pH  $\approx 4.5$ ) at 25 °C and 37 °C, respectively, and the values of  $K_{1:1} = 1480 \text{ M}^{-1}$  and  $1085 \text{ M}^{-1}$  in phosphate buffer (pH 5.5) at 25 °C and 37 °C, respectively.

### 3.2. Solid-state analyses

The DSC traces of the unprocessed flurbiprofen and SC-CO<sub>2</sub> processed flurbiprofen (Fig. 2a) showed sharp melting endotherms at 118.26 °C and 118.27 °C, respectively, confirming that SC-CO<sub>2</sub> process has not altered the crystallinity of the drug, as further substantiated by XRPD analysis (Fig. 2b). Similar results were reported by Rudrangi et al. (2015a and b) for indomethacin and olanzapine when processed with SC-CO<sub>2</sub>.

The DSC traces and X-ray powder diffractograms of flurbiprofen, Me- $\beta$ -CD and flurbiprofen-Me- $\beta$ -CD (1:1 molar) binary systems prepared by physical mixing and SC-CO<sub>2</sub> processing are presented in Figs. 3a and 3b, respectively.

The DSC trace of Me- $\beta$ -CD was characterized by a broad endothermic event between 60 °C and 120°C ascribed to its dehydration (Banchero et al. (2013); Rudrangi et al. (2015a)). The XRPD pattern of Me- $\beta$ -CD displayed two broad features at 11° and 18° indicating its predominately-amorphous nature.

The DSC traces of the binary mixtures prepared by physical mixing or SC-CO<sub>2</sub> processing (at 35 °C/100 bar, 35 °C/150 bar, 35 °C/200 bar, 45 °C/100 bar and 45 °C/150 bar) showed

endothermic peaks of reduced intensity in comparison to the unprocessed pure flurbiprofen. This was confirmed by XRPD analysis, which showed the physical mixture of flurbiprofen and Me- $\beta$ -CD to have similar diffraction pattern to that of the respective individual components but with a reduction in the intensity of drug peaks. The reduced intensity of the DSC endothermic peaks and the reduced XRPD intensities for the drug suggest an incomplete inclusion of flurbiprofen in the cavity of Me- $\beta$ -CD. The DSC trace of the binary mixtures prepared by SC-CO<sub>2</sub> processing at 35 °C/150 bar and 35 °C/200 bar showed broad endothermic peaks of drug shifted to higher temperatures, potentially attributable to the partial inclusion of the drug into cyclodextrin (Marques et al. (1990)).

A complete disappearance of the flurbiprofen melt peak was observed in the inclusion complexes prepared by SC-CO<sub>2</sub> processing at 45 °C/200 bar, indicating interactions between flurbiprofen and Me- $\beta$ -CD and the formation of inclusion complex with partially crystalline nature, as suggested by Charoenchaitrakool et al. (2002) and Rudrangi et al. (2015a and 2015b). This was supported by XRPD analysis, which showed the complexes prepared by SC-CO<sub>2</sub> processing at 35 °C/100 bar, 35 °C/150 bar, 35 °C/200 bar, 45 °C/100 bar and 45 °C/150 bar to have a considerable diminution of the diffraction peaks. A close analysis of the diffraction patterns of these samples and their comparison with that of the pure flurbiprofen showed an increase in the intensity of the drug peak at  $2\theta$  equal to 31.73°. This change may be attributed to a partial recrystallization of the drug during SC-CO<sub>2</sub> processing or the formation of the inclusion complex between flurbiprofen and Me- $\beta$ -CD as suggested by He and Li (2009) in the case of borneol and Me- $\beta$ -CD complexes.

A complete disappearance of the drug peaks but the retention of two broad features similar to that of the pure Me- $\beta$ -CD was observed in the diffraction pattern of the complexes prepared by SC-CO<sub>2</sub> processing at 45 °C/200 bar suggesting a complete complexation between flurbiprofen and Me- $\beta$ -CD.

The decrease in the degree of crystallinity of the drug was observed in the following order: SC-CO<sub>2</sub> 45°C/200 bar > SC-CO<sub>2</sub> 45°C/150 bar > SC-CO<sub>2</sub> 45°C/100 bar > SC-CO<sub>2</sub> 35°C/200 bar > SC-CO<sub>2</sub> 35°C/150 bar > SC-CO<sub>2</sub> 35°C/100 bar > Physical mixing.

### 3.3. Practical yield

The percentage practical yield of all the binary mixtures is tabulated in Table 1. It was observed that all the binary mixtures prepared by SC-CO<sub>2</sub> processing resulted in the loss of mass of the complexes to some extent. The practical yields of the binary mixtures prepared by SC-CO<sub>2</sub> processing were above 96% (w/w).

### 3.4. Evaluation of the flurbiprofen content

The percentage drug content of the binary mixtures ranged from 94.59 ± 0.31% to 97.88 ± 0.18% (Table 2), which is within the acceptable range (85-115%) provided in the United States Pharmacopoeia 29 ([USP 29, 2006](#)).

### 3.5. Dissolution studies

All the binary systems exhibited better dissolution profiles than the drug alone (Fig. 4). This enhancement can be attributed to the surfactant-like properties of cyclodextrin and the formation of inclusion complexes between flurbiprofen and Me-β-CD in the solid-state ([Lin, and Kao, 1989](#); [Guyot et al., 1995](#)). The highest improvement of the drug dissolution properties was obtained from the binary mixture prepared by SC-CO<sub>2</sub> processing at 45 °C/200 bar, followed by the mixtures processed at 45 °C/150 bar and 45 °C/100 bar.

The parameters used to characterize the dissolution curves were the percentage of flurbiprofen dissolved at 30 min, and the dissolution efficiency at 30 and 60 min. The percent drug dissolved (DP) is defined as the amount of drug dissolved at certain time, *t*, while the dissolution efficiency (DE) is defined as the area under the dissolution curve up to a certain

time,  $t$ , expressed as a percentage of the area of the rectangle described by 100 % dissolution at the same time. DE was calculated according to Khan (1975).

The percent flurbiprofen dissolved at 30 min ( $DP_{30\text{min}}$ ) and the dissolution efficiency at 30 min ( $DE_{30\text{min}}$ ) is presented in Table 3. The amount of flurbiprofen released from drug alone was very low with  $1.11 \pm 0.09\%$  dissolving at the end of 60 min, whereas the binary mixture obtained by physical mixing exhibited a significant increase in the dissolution with  $72.47 \pm 3.23\%$  of flurbiprofen dissolved at the end of 30 min.

It is also evident from the results (Fig. 4 and Table 3) that the degree of the improvement in the dissolution rate is dependent on not only the method of preparation but also the processing conditions used for the preparation of complexes. All the SC-CO<sub>2</sub> processed systems exhibited higher apparent drug dissolution than the physical mixture systems; however, considerable differences were observed in the degree of enhancement. The binary mixtures prepared by SC-CO<sub>2</sub> processing at 35 °C/100 bar exhibited a faster drug release compared to the mixtures processed at 45 °C/200 bar after 30 min ( $74.9 \pm 3.0\%$  versus  $99.4 \pm 2.3\%$ ). All the binary mixtures prepared by SC-CO<sub>2</sub> processing at 45 °C exhibited a drug release of more than 80% within the first 10 min, irrespective of the pressure employed. Both the temperature and pressure showed a positive influence on the dissolution of the flurbiprofen-Me- $\beta$ -CD inclusion complexes. At all pressures (100 bar, 150 bar and 200 bar) when the temperature was increased (from 35 °C to 45 °C), the dissolution of the complexes increased. Moreover, the increase of dissolution became more apparent at higher pressures.

The instantaneous dissolution properties offered by all the binary mixtures are attributable to the improvement in drug wettability due to the high aqueous solubility (greater than 2000 mg mL<sup>-1</sup> as reported by [Banchero et al., 2013](#)) and the superior complexing properties of Me- $\beta$ -CD ([Cirri et al., 2005](#)).

### 3.6. Docking studies

The single best pose of flurbiprofen docked in the cavity of Me- $\beta$ -CD is presented in Fig. 5. Flurbiprofen is coloured in green and the figure has a molecular surface (grey with a reddish tint in the depths), mesh surface of the hydrophilic area (cyan mesh) and the hydrophobic area (orange mesh at the mid depth of Me- $\beta$ -CD). The figure shows the binding of flurbiprofen in the cavity of Me- $\beta$ -CD through phenyl group presumably due to its hydrophobicity. Secondary hydrogen bonding exists between the oxygen atom of Me- $\beta$ -CD and the hydrogen atom of flurbiprofen. The binding affinity (GLIDE energy), van der Waals energy and docking score for the inclusion of flurbiprofen in Me- $\beta$ -CD are -20.660 kcal mol<sup>-1</sup>, -20.303 kcal mol<sup>-1</sup> and -4.259 kcal mol<sup>-1</sup>, respectively.

### Conclusions

A novel solid inclusion method for the complexation flurbiprofen into methyl- $\beta$ -cyclodextrin using supercritical carbon dioxide carrier was shown. The possibility of flurbiprofen-methyl- $\beta$ -cyclodextrin interactions of different efficiencies in the solid-state was demonstrated. Products obtained by the supercritical carbon dioxide processing exhibited the highest apparent drug dissolution followed by the physical mixture and the drug alone. All the products processed by SC-CO<sub>2</sub> at 45 °C offered more than 80% drug release within the first 10 min irrespective of the pressure employed. The method proposed involves no toxic solvent residues, thus the products obtained provide minimal side effects in humans, compared to those obtained by other techniques that require the use of organic solvents. The preliminary data suggests that the complexation of flurbiprofen with methyl- $\beta$ -cyclodextrin will lead to better therapeutic efficacy.

### Acknowledgements



The authors would like to thank Ian Slipper (University of Greenwich, UK) for his technical assistance with X-Ray powder diffraction and scanning electron microscopy studies. Madhu Battu (BITS-Pilani, Hyderabad campus, India) is thanked for his support in molecular docking studies.

## References

Al-Marzouqi, A.H., Jobe, B., Dowaidar, A., Maestrelli, F., Mura, P., 2007. Evaluation of supercritical fluid technology as preparative technique of benzocaine-cyclodextrin complexes--comparison with conventional methods. *J. Pharm. Biomed. Anal.* 43, 566–74.

DOI: <http://dx.doi.org/10.1016/j.jpba.2006.08.019>

Al-Marzouqi, A.H., Elwy, H.M., Shehadi, I., Adem, A., 2009. Physicochemical properties of antifungal drug-cyclodextrin complexes prepared by supercritical carbon dioxide and by conventional techniques. *J. Pharm. Biomed. Anal.* 49, 227–33.

DOI: <http://dx.doi.org/10.1016/j.jpba.2008.10.032>

Ambade, K.W., Jadhav, S.L., Gambhire, M.N., Kurmi, S.D., Kadam, V.J., Jadhav, K.R., 2008. Formulation and evaluation of flurbiprofen microemulsion. *Curr. Drug Deliv.* 5, 32–41.

DOI: <http://dx.doi.org/10.2174/156720108783331032>

Anderson, B.D. and Conradi, R.A., 1985. Predictive relationships in the water solubility of salts of a nonsteroidal anti-inflammatory drug. *J. Pharm. Sci.* 74, 815–820.

DOI: <http://dx.doi.org/10.1002/jps.2600740803>

Aroso, I.M., Silva, J.C., Mano, F., Ferreira, A.S.D., Dionísio, M., Sá-Nogueira, I., Barreiros, S., Reis, R.L., Paiva, A., Duarte, A.R.C., 2016. Dissolution enhancement of active pharmaceutical ingredients by therapeutic deep eutectic systems. *Eur. J. Pharm. Biopharm.* 98, 57–66. DOI: <http://dx.doi.org/10.1016/j.ejpb.2015.11.002>

Banchero, M., Ronchetti, S., Manna, L., 2013. Characterization of ketoprofen / methyl- $\beta$ -cyclodextrin complexes prepared using supercritical carbon dioxide. *J. Chem.* 2013, 8 pages.

DOI: <http://dx.doi.org/10.1155/2013/583952>

Baek, H.H., Kwon, S.Y., Rho, S.-J., Lee, W.S., Yang, H.-J., Hah, J.-M., Choi, H.-G., Kim, Y.-R., Yong, C.S., 2011. Enhanced solubility and bioavailability of flurbiprofen by cycloamylose. *Arch. Pharm. Res.* 34, 391–7. DOI: <http://dx.doi.org/10.1007/s12272-011-0306-x>

Bandi, N., Wei, W., Roberts, C.B., Kotra, L.P., Kompella, U.B., 2004. Preparation of budesonide- and indomethacin-hydroxypropyl-beta-cyclodextrin (HPBCD) complexes using a single-step, organic-solvent-free supercritical fluid process. *Eur. J. Pharm. Sci.* 23, 159–68.

DOI: <http://dx.doi.org/10.1016/j.ejps.2004.06.007>

Charman, W.N., Stella, V.J., 1991. Gastrointestinal tract as a site for drug delivery transport of lipophilic molecules by the intestinal lymphatic system. *Adv. Drug Deliv. Rev.* 7, 1–14.

DOI: [http://dx.doi.org/10.1016/0169-409X\(91\)90046-F](http://dx.doi.org/10.1016/0169-409X(91)90046-F)

Charoenchaitrakool, M., Dehghani, F., Foster, N.R., 2002. Utilization of supercritical carbon dioxide for complex formation of ibuprofen and methyl-beta-cyclodextrin. *Int. J. Pharm.* 239, 103–12. DOI: [http://dx.doi.org/10.1016/S0378-5173\(02\)00078-9](http://dx.doi.org/10.1016/S0378-5173(02)00078-9)

Cirri, M., Rangoni, C., Maestrelli, F., Corti, G., Mura, P., 2005. Development of fast-dissolving tablets of flurbiprofen-cyclodextrin complexes. *Drug Dev. Ind. Pharm.* 31, 697–707. DOI: <http://dx.doi.org/10.1080/03639040500253694>

Davies, N.M., 1995. Clinical pharmacokinetics of flurbiprofen and its enantiomers. *Clin. Pharmacokinet.* 28, 100–14. DOI: <http://dx.doi.org/10.2165/00003088-199528020-00002>

Govindarajan, R., Nagarsenker, M.S., 2005. Formulation studies and in vivo evaluation of a flurbiprofen-hydroxypropyl  $\beta$ -cyclodextrin system. *Pharm. Dev. Technol.* 10, 105–114. DOI: <http://dx.doi.org/10.1081/PDT-49687>

Gupta, G.D., Jain, S., and Jain, N.K., 1997. Formulation of an aqueous injection of flurbiprofen. *Pharmazie*, 52, 709-712.

Habib, M.J., Phan, M.T., Owusu-Ababio, G., 1998. Dissolution profiles of flurbiprofen in phospholipid solid dispersions. *Drug Dev. Ind. Pharm.* 24, 1077–1082. DOI: <http://dx.doi.org/10.3109/03639049809089952>

Heasman, P.A., Ward, A., Barrett, A.W., Seymour, R.A., Edwards, G., 1990. Flurbiprofen in human crevicular fluid analyzed by highperformance liquid chromatography. *J. Periodontal Res.* 25, 88–92. DOI: <http://dx.doi.org/10.1111/j.1600-0765.1990.tb00897.x>

He, J., Li, W., 2009. Preparation of borneol–methyl- $\beta$ -cyclodextrin inclusion complex by supercritical carbon dioxide processing. *J. Incl. Phenom. Macrocycl. Chem.* 65, 249–256. DOI: <http://dx.doi.org/10.1007/s10847-009-9575-0>

Herzfeldt, C.D., Kümmel, R., 1983. Dissociation constants, solubilities and dissolution rates of some selected nonsteroidal antiinflammatories. *Drug Dev. Ind. Pharm.* 9, 767–793. DOI: <http://dx.doi.org/10.3109/03639048309039887>

Higuchi, T., Connors, K.A., 1965. Phase Solubility Techniques. In: Reilly, C. (Ed.), *Advances in Analytical Chemistry and Instrumentation*. Wiley Interscience, New York, pp.117–212.

Jiang, Q., Yang, X., Du, P., Zhang, H., Zhang, T., 2016. Dual strategies to improve oral bioavailability of oleanolic acid: Enhancing water-solubility, permeability and inhibiting

cytochrome P450 isozymes. *Eur. J. Pharm. Biopharm.* 99, 65–72. DOI: <http://dx.doi.org/10.1016/j.ejpb.2015.11.013>

Kaialy, W., AlShafiee, M., 2015. Recent advances in the engineering of nanosized active pharmaceutical ingredients: Promises and challenges. *Adv. Colloid Interface Sci.* In Press.

Khan, K.A., 1975. The concept of dissolution efficiency: *J. Pharm. Pharmacol.* 27, 48-49. DOI: <http://dx.doi.org/10.1111/j.2042-7158.1975.tb09378.x>

Kim, C.-K., Yoon, Y.-S., Kong, J.Y., 1995. Preparation and evaluation of flurbiprofen dry elixir as a novel dosage form using a spray-drying technique. *Int. J. Pharm.* 120, 21–31. DOI: [http://dx.doi.org/10.1016/0378-5173\(94\)00375-F](http://dx.doi.org/10.1016/0378-5173(94)00375-F)

Kompella, U. B.; Koushik, K., 2001. Preparation of drug delivery systems using supercritical fluid technology. *Crit. Rev. Ther. Drug Carr. Syst.* 18 (2), 173–199. DOI: <http://dx.doi.org/10.1615/CritRevTherDrugCarrierSyst.v18.i2.20>

Lee, L.Y., Wang, C.H., Smith, K.A., 2008. Supercritical antisolvent production of biodegradable micro- and nanoparticles for controlled delivery of paclitaxel. *J. Control. Release* 125, 96–106. DOI: <http://dx.doi.org/10.1016/j.jconrel.2007.10.002>

Li, D., Han, M., Balakrishnan, P., Yan, Y., Oh, D., Joe, J., Seo, Y., Kim, J., Park, S., Yong, C., Choi, H.-G., 2010. Enhanced oral bioavailability of flurbiprofen by combined use of micelle solution and inclusion compound. *Arch. Pharm. Res.* 33, 95–101. DOI: <http://dx.doi.org/10.1007/s12272-010-2231-9>

Muraoka, A., Tokumura, T., Machida, Y., 2004. Evaluation of the bioavailability of flurbiprofen and its beta-cyclodextrin inclusion complex in four different doses upon oral administration to rats. *Eur. J. Pharm. Biopharm.* 58, 667–71. DOI: <http://dx.doi.org/10.1016/j.ejpb.2004.03.030>

Oh, D.H., Park, Y.-J., Kang, J.H., Yong, C.S., Choi, H.-G., 2010. Physicochemical characterization and in vivo evaluation of flurbiprofen-loaded solid dispersion without crystalline change. *Drug Deliv.* 18, 46–53.

DOI: <http://dx.doi.org/10.3109/10717544.2010.509365>

Park, K.-M., Gao, Z.-G., Kim, C.-K., 1997. Assay of flurbiprofen in rat plasma using HPLC with fluorescence detection. *J. Liq. Chromatogr. Relat. Technol.* 20, 1849–1855.

DOI: <http://dx.doi.org/10.1080/10826079708005547>

Rudrangi, S.R.S., Rudrangi, R., Bontha, V.K., Rudrangi, S., 2011. Prodrugs: Legendary 'Magic Bullets' for delivery of therapeutics. *Int. J. Pharm. Sci. Rev. Res.* 11 (2), 7-12.

Bontha, V.K., Pabbu, B.B., Prasad, G., Rudrangi, S.R.S., Chilukala, S., Rudrangi, S., 2012. Formulation and evaluation of isradipine buccal tablets. *Res. J. Pharm. Tech.* 5(9), 1187-1196

Rudrangi, S.R.S., Bhomia, R., Trivedi, V., Vine, G.J., Mitchell, J.C., Alexander, B.D., Wicks, S.R., 2015a. Influence of the preparation method on the physicochemical properties of indomethacin and methyl- $\beta$ -cyclodextrin complexes. *Int. J. Pharm.* 479(2), 381–390. DOI:

<http://dx.doi.org/10.1016/j.ijpharm.2015.01.010>

Rudrangi, S.R.S., Trivedi, V., Mitchell, J.C., Alexander, B.D., Wicks, S.R., 2015b. Preparation of olanzapine and methyl- $\beta$ -cyclodextrin complexes using a single-step, organic solvent-free supercritical fluid process: An approach to enhance the solubility and dissolution properties *Int. J. Pharm.* 494(1), 408-416.

DOI: <http://dx.doi.org/10.1016/j.ijpharm.2015.08.062>

Sunkara, G., Kompella, U.B., 2002. Drug delivery applications of supercritical fluid technology. *Drug. Del. Technol.* 2, 44–50.

Takagi, T., Ramachandran, C., Bermejo, M., Yamashita, S., Yu, L.X., Amidon, G.L., 2006. A Provisional Biopharmaceutical Classification of the Top 200 Oral Drug Products in the United States, Great Britain, Spain, and Japan. *Mol. Pharm.* 3, 631–643. [DOI: http://dx.doi.org/10.1021/mp0600182](http://dx.doi.org/10.1021/mp0600182)

The United States Pharmacopoeia 29 (USP 29), 2006. United States Pharmacopoeial Convention Inc., Rockville, 958-959.

Tokumura, T., Muraoka, A., Machida, Y., 2009. Improvement of oral bioavailability of flurbiprofen from flurbiprofen/beta-cyclodextrin inclusion complex by action of cinnarizine. *Eur. J. Pharm. Biopharm.* 73, 202–204. [DOI: http://dx.doi.org/10.1016/j.ejpb.2009.04.018](http://dx.doi.org/10.1016/j.ejpb.2009.04.018)

Türk, M., Upper, G., Steurethaler, M., Hussein, K., Wahl, M.A., 2007. Complex formation of Ibuprofen and  $\beta$ -Cyclodextrin by controlled particle deposition (CPD) using SC-CO<sub>2</sub>. *J. Supercrit. Fluids* 39, 435–443. [DOI: http://dx.doi.org/10.1016/j.supflu.2006.02.009](http://dx.doi.org/10.1016/j.supflu.2006.02.009)

Varma, M.M., Pandi, J.K., 2005. Dissolution, solubility, XRD, and DSC studies on flurbiprofen-nicotinamide solid dispersions. *Drug Dev. Ind. Pharm.* 31, 417–423. [DOI: http://dx.doi.org/10.1080/03639040500214613](http://dx.doi.org/10.1080/03639040500214613)

Vega, E., Egea, M.A., Garduño-Ramírez, M.L., García, M.L., Sánchez, E., Espina, M., Calpena, A.C., 2013. Flurbiprofen PLGA-PEG nanospheres: Role of hydroxy- $\beta$ -cyclodextrin on ex vivo human skin permeation and in vivo topical anti-inflammatory efficacy. *Colloids Surfaces B Biointerfaces* 110, 339–346.

[DOI: http://dx.doi.org/10.1016/j.colsurfb.2013.04.045](http://dx.doi.org/10.1016/j.colsurfb.2013.04.045)

Weerapol, Y., Limmatvapirat, S., Jansakul, C., Takeuchi, H., Sriamornsak, P., 2015. Enhanced dissolution and oral bioavailability of nifedipine by spontaneous emulsifying

powders: Effect of solid carriers and dietary state. *Eur. J. Pharm. Biopharm.* 91, 25–34. DOI: <http://dx.doi.org/10.1016/j.ejpb.2015.01.011>

York, P., 1999. Strategies for particle design using supercritical fluid technologies. *Pharm. Sci. Technol. Today* 2, 430–440. DOI: [http://dx.doi.org/10.1016/S1461-5347\(99\)00209-6](http://dx.doi.org/10.1016/S1461-5347(99)00209-6)