

Influence of continuous exposure to gaseous ozone on the quality of red bell peppers, cucumbers and zucchini

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ABSTRACT

The effect of continuous exposure to ozone on quality changes during the storage of red bell peppers, cucumbers and zucchini was investigated. Peppers were stored at 14 °C and were exposed to ozone at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$, while cucumbers and zucchini were stored at 12 and 8 °C, respectively and exposed to ozone at 0.1 $\mu\text{mol mol}^{-1}$. The content of fructose (2.75 g/100 g FW) and glucose (2.00 g/100 g FW) in red bell peppers exposed to ozone at 0.1 $\mu\text{mol mol}^{-1}$ was increased by 8 and 7%, respectively when compared to controls. Continuous exposure to ozone at 0.3 $\mu\text{mol mol}^{-1}$, on the other hand, had no effect on fructose (2.52 g/100 g FW) and glucose (1.88 g/100 g FW) content. The content of vitamin C was significantly enhanced in red bell peppers exposed to ozone at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$ after 7 days of storage, however, this effect was not maintained. After 14 days, vitamin C content in peppers exposed to ozone at 0.1 $\mu\text{mol mol}^{-1}$ was not significantly different from the control, whereas it was reduced at 0.3 $\mu\text{mol mol}^{-1}$. Total phenolics content was increased in peppers exposed to ozone at 0.1 $\mu\text{mol mol}^{-1}$, but was unaffected at 0.3 $\mu\text{mol mol}^{-1}$. Continuous exposure of red bell peppers to ozone at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$ had no significant effect on weight loss, texture and colour. In cucumbers and zucchini, continuous exposure to ozone at 0.1 $\mu\text{mol mol}^{-1}$ reduced weight loss by more than 40% and improved texture maintenance, while having no significant effect on their biochemistry. The findings from this study suggest that continuous exposure to ozone at 0.1 $\mu\text{mol mol}^{-1}$ is a promising method for shelf-life extension of cucumbers and zucchini. Even though in red bell peppers continuously exposed to ozone at 0.1 $\mu\text{mol mol}^{-1}$ sugars and phenolics content was increased, further work is still needed to better understand the exact mechanism of ozone action and its potential for the industrial use.

Keywords:

Fresh produce

Quality evaluation

Storage

1. Introduction

The economic value of trade in fresh produce is constantly growing, due to increasing consumer demand. Consumers care more and more about what they eat and fresh produce has been recognised as a healthy food, for example being rich in antioxidants (Llorach et al., 2008; Alothman et al., 2010; Yeoh et al., 2014). The shelf-life of fresh produce, however, is shorter than other food products, and is determined by initial quality at

harvest (Clarkson et al., 2003; Zhang et al., 2007) and subsequent storage conditions (Nunes et al., 2009). New techniques for reducing undesired microbial contamination, spoilage and decay, as well as maintaining the product's visual, textural and nutritional quality are required at all steps of the production and distribution chain.

Treatment with ozone (O₃) is currently being explored as a practical method to reduce/eliminate microorganisms present in food (Khadre et al., 2001; Guzel-Seydim et al., 2004). Ozone is a well-known strong oxidizing agent that has been used by the fresh produce industry as an antimicrobial agent for a number of years and has been generally recognised as safe (GRAS). The use of ozone by the fresh produce industry is a good alternative to chemical treatments, such as the use of chlorine as it leaves no chemical residues. Recently, there has been an increasing interest in the use of ozone as a postharvest treatment of fruit and vegetables (Horvitz and Cantalejo, 2014). Only those treatments that reduce microbial contamination and extend the shelf-life of the product without having an adverse effect on the product's visual, textural and nutritional quality (Allende et al., 2008) can be recommended and subsequently incorporated into the supply chain.

A number of studies (Ketteringham et al., 2006; Alexandre et al., 2011; Horvitz and Cantalejo, 2012; Alexopoulos et al., 2013) investigated the efficiency of ozone in reducing microbial counts on peppers, however, only one study (Horvitz and Cantalejo, 2012) assessed physicochemical properties of the produce. Microbial contamination of peppers can be reduced by applying ozone in either gaseous (Horvitz and Cantalejo, 2012) or aqueous (Alexandre et al., 2011; Alexopoulos et al., 2013) form. Ketteringham et al. (2006), however, did not observe reductions in microbial counts in fresh-cut peppers washed with ozonated water at 0.30-0.35, 0.38-0.45 and 3.85-3.95 $\mu\text{mol mol}^{-1}$ for 20 s to 30 min, and suggested that this could be due to cut surfaces that promoted leaching of organic matter, thus providing a higher concentration of organic matter to react with ozone, thereby reducing the concentration of ozone available to act as an antimicrobial agent. Thus, Ketteringham et al. (2006) suggested treating whole rather than pre-cut peppers. Interestingly, ozonated water at 1 $\mu\text{mol mol}^{-1}$ applied for 3-5 min, has recently been found to be efficient in reducing mesophilic and psychrotrophic bacteria, yeast and mould counts on fresh-cut peppers (Horvitz and Cantalejo, 2012). Furthermore, Horvitz and Cantalejo (2012) reported that gaseous ozone at 0.7 $\mu\text{mol mol}^{-1}$ applied for 1-5 min prior to storage was even more efficient as a sanitizer when compared with aqueous form at 1 $\mu\text{mol mol}^{-1}$ applied for the same time. These findings highlight the need to focus on the effects of gaseous ozone on the quality of red bell peppers. Fruit and vegetables can be treated either with high ozone concentration prior to storage (Yeoh et al., 2014) or they might be continuously/intermittently exposed to lower ozone concentration

during storage (Aguayo et al., 2006; Tzortzakis et al., 2007b). There are no reports on continuous exposure of red bell peppers to low concentrations of ozone.

On the other hand, the availability of information regarding the effects of ozone exposure on the quality of cucumbers is limited (Skog and Chu, 2001), while to our knowledge, there has not been any research dealing with effects of ozone exposure on zucchini.

Exposure of fresh produce to ozone is expected to induce production of reactive oxygen species (ROS) (Kangasjarvi et al., 2005); too much stress, however, leads to cell death, evidenced by discolouration and loss of texture. Thus, the dose of ozone has to be appropriately adjusted for each commodity (Forney et al., 2003). For safety reasons, within the fresh produce industry, it is important to keep the levels of ozone low; the recommended limit set for humans by the US Occupational Health and Safety Administration is $0.1 \mu\text{mol mol}^{-1}$ averaged over an 8 hour shift. Ozone can also cause damage to equipment by, for example causing cracks in rubber. Thus, the aim of this study was to determine the effect of continuous exposure to low concentrations of ozone ($0.1\text{-}0.3 \mu\text{mol mol}^{-1}$) on quality changes during the storage of red bell peppers, cucumbers and zucchini.

2. Materials and methods

2.1. Plant material and handling

Red bell peppers (*Capsicum annuum* L.) variety Ferrari (commercially mature but not fully ripe; 75-85 mm in diameter) and mature cucumbers (*Cucumis sativus* L.) variety High Jack (~30 cm long) were supplied by a commercial greenhouse and pack-house facility, Thanet Earth, Kent, UK, whereas zucchini (*Cucurbita pepo* L.) variety Prometheus (small; 12-16 cm long) were supplied by Mack Multiples Ltd, Kent, UK. On arrival, all fruit were graded to be free from visible defects.

For experiment 1, that addressed the effect of ozone exposure on quality changes during the storage of red bell peppers, fruit were placed for 14 d in six (2 containers per treatment) 30 L sealable plastic containers (24 fruit per treatment) supplied with humidified air (RH, $92\pm 2\%$) at constant air flow of $0.05 \text{ m}^3 \text{ h}^{-1}$ as a continuous, flow-through system at $14\pm 1 \text{ }^\circ\text{C}$, as recorded using temperature and humidity loggers (Lascar Electronics Ltd, UK). Ozone was supplied at approximately 0.1 ± 0.015 and $0.3\pm 0.030 \mu\text{mol mol}^{-1}$, using ozone generators (Onnic International, UK) placed within the containers close to air inlet. Air was circulated inside the box to ensure even distribution of ozone. Ozone concentration was monitored periodically, on the sampling day before taking the produce out from containers for subsequent assessment, with an L-106 Ozone Monitor (2B Technologies, US). Produce quality (weight loss, texture, colour, sugars, soluble solids and pH, ascorbic acid and total phenolics) was assessed on arrival and after 7 and 14 d of storage, respectively.

For experiment 2, that addressed the effect of ozone exposure on quality changes during the storage of cucumbers and zucchini, fruit were placed for 17 d in sixteen (4 containers per treatment) 30 L sealable plastic containers (48 fruit per treatment) supplied with humidified air (RH, 90±2%) at a constant air flow of 0.05 m³ h⁻¹ as a continuous, flow-through system at 12±1 °C (cucumbers) and 8±1 °C (zucchini), respectively. Ozone at 0.3±0.030 μmol mol⁻¹ was found injurious to the fruit, thus in experiment 2 only the lower dose at approximately 0.1±0.015 μmol mol⁻¹ was used. Produce quality (weight loss, texture, colour, ascorbic acid and total phenolics), however, was assessed more frequently, i.e. on arrival and after 6, 10, 13 and 17 d of storage, respectively.

Storage conditions for both experiments were advised by fruit suppliers to simulate the conditions that produce is facing at their facilities, so that the findings from this research could have a practical value for them.

2.2. Measurements

2.2.1. Weight loss

All fruit were labelled and weighed on arrival (day 0). Weight loss (%) was determined by comparing the weight of each fruit on the sampling day with their initial weight determined on day 0.

2.2.2. Texture analysis

Firmness of red bell peppers was determined following the method of Vega-Galvez et al. (2009) with some modifications. Fruit firmness was determined (4 measurements per fruit) using a TA.XT plus Texture Analyser (Stable Micro Systems, UK) equipped with a 2-mm diameter probe (puncture test) and a 0.05 kN load cell. The probe was driven 5.0 mm at a speed of 1.7 mm s⁻¹ and the maximum force (N) was recorded.

Firmness of cucumbers and zucchini was determined following the method of Hurr et al. (2013) with some modifications. Fruit firmness was determined (5 measurements per fruit) using a TA.XT plus Texture Analyser (Stable Micro Systems, UK) equipped with a convex-tip probe; 8-mm diameter for whole fruit firmness and 2-mm diameter for mesocarp firmness, and a 0.05 kN load cell. The probe was driven 2.5 mm at a speed of 0.83 mm s⁻¹ and the maximum force (N) was recorded.

2.2.3. Colour

Skin colour measurements were taken using a Minolta CR-400 chroma meter (Minolta, Japan) with an 8 mm diameter measuring head and a C illuminant calibrated with manufacturer's standard white plate. Colour changes were quantified for 12 fruit from each sample (5 measurements per pepper, 3 measurements per cucumber and zucchini) in the L^* , a^* and b^* colour space (Abbott, 1999). Hue angle (H°) was then calculated as $H^\circ = \tan^{-1}(b^*/a^*)$, when a^* and b^* were >0 and $H^\circ = 180 + \tan^{-1}(b^*/a^*)$, when a^* was <0 and b^* was >0 .

2.2.4. Soluble solids and pH

Soluble solids content (SSC) was measured using an eclipse handheld refractometer (Bellingham & Stanley Ltd, UK) and expressed in °Brix. The pH was measured using a Jenway 3510 pH meter (Bibby Scientific Ltd, UK).

2.2.5. Sugar analyses

Freeze-dried pepper samples were ground into powder using a mortar and pestle. Extracts for sugars determination were prepared as described elsewhere (Gine Bordonaba and Terry, 2010) with some modifications. Briefly, aliquots of 50 mg were extracted with 1 ml of ethanol (Fisher Scientific Ltd, UK) for 2 h in a shaking water bath at 70 °C. Samples were centrifuged at 10,000 x g for 4 minutes and then filtered through 0.45 µl syringe filters (Chromacol Ltd, UK). The volume of 10 µl was injected into an Agilent 1200 series HPLC (Agilent, UK) with a Zorbax Carbohydrate Analysis column (150 mm x 4.6 mm x 5 µm) (Agilent, UK) and Zorbax NH₂ guard column (Agilent, UK) at a flow rate of 1.0 ml min⁻¹. The mobile phase consisted of 75% of acetonitrile (Fisher Scientific Ltd, UK) and 25% water. The concentration of fructose, glucose and sucrose was determined according to external fructose, glucose and sucrose standards (Sigma-Aldrich, UK).

2.2.6. Total ascorbic acid extraction and determination

Ascorbic acid (AsA) was extracted and analysed using a method described by Bergquist et al. (2006) with some modifications. Samples were extracted with cold (4 °C) 3.0% (30 g/l w/v in H₂O) *meta*-phosphoric acid (HPO₃) (Sigma-Aldrich, UK). Samples were homogenized with a Janke & Kunkel IKA Labor Technik ultraturrax T25 homogenizer (IKA, Germany). The extracts were centrifuged at 4,000 x g for 40 min at 4 °C. Supernatants were filtered through 0.45 µl syringe filters (Chromacol Ltd, UK) and 1.0 ml was collected in Eppendorf tubes. Following filtration, extracts were microfuged at 9,300 x g for 5 min. Finally, 500 µl was transferred into HPLC vials for AsA determination. Another 500 µl was transferred into new 1.5 ml Eppendorf tubes and mixed thoroughly, with an equal volume of 1% (11 mg/ml w/v in 1 M K₂HPO₄/H₂O (1/4, v/v)) DTT solution (DL-Dithiothreitol) (Sigma-Aldrich, UK). These samples were left for 40 min at room temperature (20±1.0 °C), and then microfuged at 9,300 x g for 5 min. Samples were transferred into HPLC vials for total AsA (AsA + DHA) determination. Samples were analysed using an Agilent 1100 HPLC (Agilent, UK) with a Luna 5 µm NH₂ 100 A column (250 mm × 4.6 mm) (Phenomenex, UK) at a flow rate of 1.2 ml min⁻¹. The mobile phase consisted of 25% 15 mmol l⁻¹ (1.725 g/l w/v in H₂O) of NH₄H₂PO₄ (mono ammonium phosphate) (Sigma-Aldrich, UK) and 75% of acetonitrile (Fisher Scientific Ltd, UK); pH was adjusted to 3.9 with 1 M *ortho*-phosphoric acid (H₃PO₄) (Acros Organics, UK). The concentration of AsA and DHA was determined

according to external AsA standards (Sigma-Aldrich, UK) of 10, 25, 50 and 100 $\mu\text{mol mol}^{-1}$. The volume of 20 μl of each sample was analysed in this process.

2.2.7. Total phenolics determination

Total phenolic content (TPC) of the fruit extracts was determined as gallic acid equivalent (GAE) using the Folin–Ciocalteu assay, which was described by Singleton and Rossi (1965). Each sample was extracted with 80% methanol (Fisher Scientific Ltd, UK). A 20 μl aliquot of extract was diluted with 1.58 ml deionised water, and then was mixed with 100 μl of Folin & Ciocalteu's phenol reagent (Sigma-Aldrich, UK). After standing for 4 min at room temperature, 300 μl of the sodium carbonate (Sigma-Aldrich, UK) solution (200 g/L w/v) was added. The solutions were mixed and allowed to stand for 2 h in the dark at room temperature (20.0 ± 1.0 °C). Next, the absorbance was measured at 765 nm using a CE 9200 double beam UV/VIS spectrophotometer (CECIL Instruments Limited, UK). A calibration curve was prepared using a standard solution of gallic acid (Sigma-Aldrich, UK) of 50, 100, 150, 250, and 500 mg/L.

2.2.8. Disease incidence

On the sampling days, red bell peppers, cucumbers and zucchini were assessed for signs of rotting, by giving them a score (0 or 1 – no/ signs of rotting, respectively). Disease incidence (DI) was expressed as the proportion (%) of fruit showing signs of rotting out of the total number of fruit in each treatment.

2.2.9. Ethylene concentration

Air samples (~0.5 l) from the containers with fresh produce were taken in duplicates for each container using SKC Quality Sample Bags and SKC Vac-U-Chamber (SKC Ltd., UK). Ethylene concentration was determined by injecting a 1 ml sample of headspace into a gas chromatograph (ATI-Unicam 610 series) fitted with a flame ionization detector (FID) set at 250 °C and a 1 m long, 6 mm OD glass column packed with 100/120 mesh.

2.3. Statistical analysis

Data are presented as mean values from a fully randomised design. The significance of main effect was established using ANOVA. Tukey's test was used to compare individual treatment values. All statistical analyses were performed using GenStat 11th Edition (Payne et al., 2008) software (VSN International Ltd, UK).

3. Results and discussion

3.1. Weight loss

As expected, red bell peppers lost weight progressively over the storage period; after 7 d of storage at 14 °C, weight loss on average was in the range of 1-2% (1.4, 1.0 and 1.6% in control and peppers exposed to

ozone at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$, respectively) and increased to 3-4% (3.8, 3.6 and 3.5% in control and peppers exposed to ozone at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$, respectively) after 14 d, respectively. No significant difference, however, was found between control peppers and those exposed to ozone at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$. These findings are in agreement with others, who observed that in peppers exposed to relatively low concentration of ozone weight loss was unaffected (Horvitz and Cantalejo, 2012) with similar response being observed in tomatoes (Tzortzakis et al., 2007a).

On the other hand, a significantly ($P < 0.05$) higher weight loss, was observed in air-stored (control) cucumbers when compared with their ozone-exposed (0.1 $\mu\text{mol mol}^{-1}$) counterparts. This difference was already apparent after 6 d of storage at 12 °C, when weight loss in control cucumbers reached 3.3%, whereas it was only 1.5% in their ozone-exposed counterparts. At the end of the storage period, weight loss in ozone-exposed cucumbers did not exceed 3.0%, while the control samples lost nearly 6.0% of their weight, indicating that ozone exposure prevented water loss from the fruit. Similar results were observed in zucchini, where at the end of the storage period, in ozone-exposed samples weight loss did not exceed 5.0%, whereas in control samples, with visible signs of shrivelling, it reached 12.0%. These findings are notable, since in the majority of studies (Fomey et al., 2003; Horvitz and Cantalejo, 2012) ozone exposure had no effect on weight loss of produce. However, in a recent study of Ali et al. (2014), exposure to ozone at 1.5 to 5 $\mu\text{mol mol}^{-1}$ for 96 h resulted in reduced weight loss of papaya fruit. These authors suggested that this positive response might be related to the thick cuticle of the papaya fruit which prevented the damage of epidermal tissues by ozone action. Kechinsky et al. (2012) studied the effects of ozone on the epidermis of the papaya fruit with a use of scanning electron microscope images. They reported that ozone treatment at 4 $\mu\text{mol mol}^{-1}$ for 1 and 2 min did not affect the fruit surface. The mechanism of ozone action may involve its effect on stomata, which have previously been found to be closed in response to ozone in cucumber (Agrawal et al., 1993) and spinach (Calatayud et al., 2004) leaves. The amount of water loss by a fruit during storage, however, may also vary due to cuticle thickness, and this probably explains the higher weight loss observed in zucchini when compared with cucumbers.

Interestingly, Skog and Chu (2001) did not observe any effect on weight loss when cucumbers were exposed to ozone at 0.04 $\mu\text{mol mol}^{-1}$ during storage for 12 d at 10 °C. This different response might be due to different dose of ozone being used in their study or the fact that they stored cucumbers in the presence of ethylene at 1.5-2.0 $\mu\text{mol mol}^{-1}$. Exposure to ethylene has been reported to reduce the quality of cucumbers (Hurr et al., 2013) leading to softening of the fruit.

3.2. Texture

Texture loss during storage is a serious problem because it reduces marketability of the product. Red bell peppers are highly perishable and are not suitable for long term storage. Thus, it is not surprising that texture loss from 8.85 N to 6.96, 7.30 and 6.86 N in control and peppers exposed to 0.1 and 0.3 $\mu\text{mol mol}^{-1}$ ozone respectively was observed after 14 d of storage at 14 °C. Texture loss was partly associated with the water loss. The lowest texture was observed in peppers exposed to ozone at 0.3 $\mu\text{mol mol}^{-1}$ which were slightly but not significantly softer when compared with other samples. In agreement with the study of Horvitz and Cantalejo (2012), ozone exposure at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$ had no significant effect on texture maintenance in red bell peppers.

The firmness of both cucumbers and zucchini (Fig. 1), on the other hand, was found to be significantly ($P<0.05$) better maintained in fruit samples exposed to ozone at 0.1 $\mu\text{mol mol}^{-1}$; the effect being more pronounced in zucchini (Fig. 1B), probably due to higher water loss of zucchini control samples. Mesocarp firmness was not affected by the treatment. These results are not surprising, as several studies have already reported better firmness retention in ozone exposed fruit, e.g. in cucumbers continuously exposed to ozone at 0.04 $\mu\text{mol mol}^{-1}$ (Skog and Chu, 2001), in papaya exposed to ozone at 1.5-3.5 $\mu\text{mol mol}^{-1}$ for 96 h (Ali et al., 2014) and tomatoes (Aguayo et al., 2006; Tzortzakis et al., 2007a; Rodoni et al., 2010), where softening of the fruit, associated with ripening, was delayed in ozone-exposed samples. Rodoni et al. (2010) conducted analyses of the cell walls, and found a decreased activity of pectin methylesterase (PME) in ozone-exposed tomato fruit. These authors suggested that delayed fruit softening was related to reduced solubilisation and depolymerisation of pectin polysaccharides.

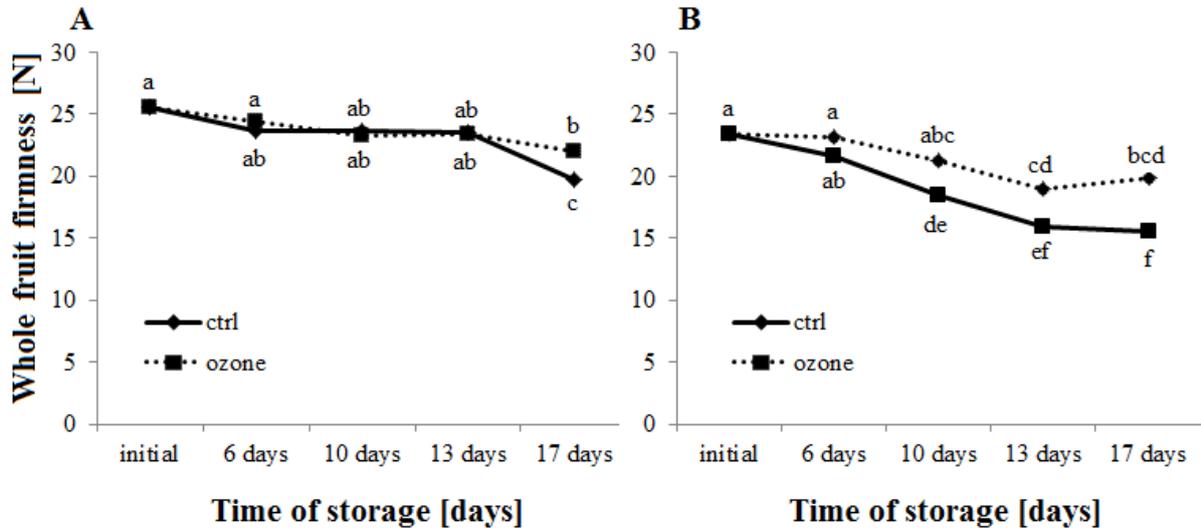


Fig. 1. Effect of continuous exposure to ozone at $100 \mu\text{mol mol}^{-1}$ on firmness [N] maintenance during the storage of cucumbers (A) and zucchini (B) for 17 d at 12 and 8 °C, respectively. Data represent mean values from 12 replicates. Different letters indicate that values are significantly different ($P < 0.05$).

It is apparent from this study that in those commodities, where exposure to ozone can significantly reduce water loss during storage, texture maintenance would be improved. Ozone dose (ozone concentration x time of exposure), however, has to be properly adjusted, so that it is not injurious to the produce.

3.3. Colour

The colour of the epidermis of the red bell peppers at harvest was orange/red and corresponded to a hue angle (H°) of 40.03. This value, together with b^* value declined during 14 d of storage at 14 °C, indicating that the skin colour changed to dark red. No difference in H° , L^* , a^* and b^* values, however, was observed between control and ozone-treated samples throughout the storage period (Table 1). These findings are similar to those reported by Horvitz and Cantalejo (2012), who reported that ozone treatment at $0.7 \mu\text{mol mol}^{-1}$ for up to 5 min had no effect on colour of minimally processed peppers. This finding suggest that continuous exposure to low doses of ozone, i.e. $0.1\text{-}0.3 \mu\text{mol mol}^{-1}$, would neither bleach the pigments nor be injurious to the produce.

Table 1

Effect of ozone exposure at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$ on colour characteristics during the storage of peppers for 14 d at 14 °C. Data represent mean values from 12 replicates. Different letters indicate that values are significantly different ($P<0.05$).

Time of storage	Treatment	L^* (lightness)	a^* (redness)	b^* (yellowness)	H° (hue angle)
Initial		37.71 a	26.14 c	22.06 a	40.03 a
	ctrl	33.05 c	29.17 a	21.30 a	36.13 b
Day 7	0.1 $\mu\text{mol mol}^{-1}$	33.22 bc	28.65 ab	20.87 a	35.37 b
	0.3 $\mu\text{mol mol}^{-1}$	33.60 bc	29.32 a	21.72 a	36.50 b
	ctrl	34.26 bc	27.14 bc	17.13 b	32.21 c
Day 14	0.1 $\mu\text{mol mol}^{-1}$	33.76 bc	27.50 bc	17.43 b	32.21 c
	0.3 $\mu\text{mol mol}^{-1}$	34.37 b	27.25 bc	17.74 b	33.02 c

The colour of the fruit epidermis at harvest was dark green and corresponded to a hue angle (H°) of 125.93 and 128.00 in cucumbers and zucchini, respectively. This value declined significantly in cucumbers during 17 d of storage at 12 °C due to increase in skin yellowness (b^* value) indicating that skin colour changed from dark to light green. In zucchini, on the other hand, no significant difference in hue angle was found over the storage period, suggesting that colour changes were less pronounced or more difficult to detect in this commodity, probably due to non-uniform colour of the skin. No difference in H° , L^* , a^* and b^* values, however, was observed between control and ozone-treated cucumbers and zucchini samples throughout the storage period (Table 2), with the exception of zucchini after 17 d of storage at 8 °C, where control fruit became significantly ($P<0.05$) more yellow. These findings are in agreement with Skog and Chu (2001), who did not observe any colour alteration in ozone exposed cucumbers. The yellowing of control zucchini at the end of storage, on the other hand, may be explained by progressing yellowing and decay, indicating senescence, associated with substantial water loss.

Table 2

Effect of ozone exposure at $0.1 \mu\text{mol mol}^{-1}$ on colour characteristics during the storage of cucumbers and zucchini for 17 d at 12 and 8 °C, respectively. Data represent mean values from 12 replicates. Different letters (for each commodity) indicate that values are significantly different ($P<0.05$).

Time of storage	Treatment	<i>L*</i> (lightness)	<i>a*</i> (greenness)	<i>b*</i> (yellowness)	<i>H*</i> (hue angle)
<i>cucumbers</i>					
Initial		31.82 e	-11.11 a	15.35 e	125.93 a
Day 6	ctrl	36.44 d	-12.76 ab	18.08 d	125.38 a
	$0.1 \mu\text{mol mol}^{-1}$	36.08 d	-12.12 a	16.88 de	126.01 a
Day 10	ctrl	36.25 d	-12.24 a	16.99 de	125.89 a
	$0.1 \mu\text{mol mol}^{-1}$	38.18 cd	-13.40 abc	19.10 cd	125.22 ab
Day 13	ctrl	39.77 bc	-13.98 abc	21.34 bc	123.38 c
	$0.1 \mu\text{mol mol}^{-1}$	39.41 bc	-13.91 abc	21.31 bc	123.55 bc
Day 17	ctrl	41.03 ab	-14.28 bc	23.96 ab	121.28 d
	$0.1 \mu\text{mol mol}^{-1}$	43.00 a	-15.03 c	25.53 a	120.88 d
<i>zucchini</i>					
Initial		27.34 a	-3.75 a	4.82 b	128.00 a
Day 6	ctrl	28.68 a	-4.67 a	6.39 b	126.20 a
	$0.1 \mu\text{mol mol}^{-1}$	28.48 a	-4.55 a	6.27 b	126.00 a
Day 10	ctrl	29.25 a	-5.28 a	7.33 ab	126.30 a
	$0.1 \mu\text{mol mol}^{-1}$	27.80 a	-4.64 a	6.53 b	125.60 a
Day 13	ctrl	28.64 a	-4.31 a	5.95 b	126.30 a
	$0.1 \mu\text{mol mol}^{-1}$	29.42 a	-5.18 a	7.19 ab	125.50 a
Day 17	ctrl	30.93 a	-6.30 a	9.65 a	124.30 a
	$0.1 \mu\text{mol mol}^{-1}$	28.61 a	-4.64 a	6.55 b	125.50 a

3.4. Physicochemical properties

The taste of the product can be affected by a number of factors, including sugar content and composition, acidity, organic acids and texture-related mouth feel (Eggink et al., 2012; Piombino et al., 2013). The level of soluble solids is associated with sugar content and fruit maturity. It generally increases during fruit ripening, and starts to decline when fruit are overripe. The amount of soluble solids increased slightly during the storage of red bell peppers. After 14 d of storage the lowest level of soluble solids was observed in samples exposed to ozone at $0.1 \mu\text{mol mol}^{-1}$. Even though, no significant difference was found between the samples, these findings may suggest that there is a potential to slow the ripening of red bell peppers if the dose of ozone is properly adjusted. Fruit ripening has already been found to be delayed in kiwi continuously exposed to ozone at $0.3 \mu\text{mol mol}^{-1}$ (Minas et al., 2012) and papaya fruit exposed to ozone at $1.5\text{-}3.5 \mu\text{mol mol}^{-1}$ for 96 h (Ali et al., 2014) but not when the concentration of ozone was higher, inhibiting the development of full ripe colour and

resulting in tissue damage due to its strong oxidizing activity and thus supporting the growth of fungal pathogens. This underlines the fact that only at certain levels of ozone shelf-life may be extended.

Similar to the study of Horvitz and Cantalejo (2012), ozone exposure had no significant effect on the pH of pepper juice (Table 3). This is not surprising, since even in those studies that observed some changes in organic acids in produce exposed to ozone (Aguayo et al., 2006; Barboni et al., 2010) the acidity of the product was not affected.

Table 3

Effect of ozone exposure at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$ on physicochemical properties of red bell peppers during the storage for 14 d at 14 °C. Data represent mean values from 12 replicates. Different letters indicate that values are significantly different ($P<0.05$).

Time of storage	Treatment	pH	Soluble solids (*Brix)	Fructose (g/100 g FW)	Glucose (g/100 g FW)	Sucrose (g/100 g FW)
Day 7	ctrl	5.02 a	5.82 ab	2.67 ab	1.97 ab	0.06 a
	0.1 $\mu\text{mol mol}^{-1}$	5.08 a	5.60 ab	2.70 ab	2.07 a	0.07 a
	0.3 $\mu\text{mol mol}^{-1}$	5.06 a	5.43 b	2.60 ab	1.98 ab	0.06 a
Day 14	ctrl	5.03 a	6.12 a	2.54 b	1.87 b	0.06 a
	0.1 $\mu\text{mol mol}^{-1}$	5.05 a	5.85 ab	2.75 a	2.00 ab	0.06 a
	0.3 $\mu\text{mol mol}^{-1}$	5.03 a	5.97 ab	2.52 b	1.88 b	0.04 b

Exposure of red bell peppers to ozone at the concentration of 0.1 $\mu\text{mol mol}^{-1}$ led to increased content of fructose and glucose, whereas ozone at 0.3 $\mu\text{mol mol}^{-1}$ had no effect on fructose and glucose but reduced the content of sucrose (Table 3). These results are similar to those of Aguayo et al. (2006), who observed higher fructose and glucose content in tomatoes exposed to ozone applied cyclically at 4 $\mu\text{mol mol}^{-1}$ for 30 min every 3 h. The sugar loss in control samples and those exposed to ozone at 0.3 $\mu\text{mol mol}^{-1}$ may be associated with increased respiration. Respiration rate (consumption of O_2 and production of CO_2), which is a measure of physiological activity (Pirovani et al., 1998) increases in response to tissue damage and as demonstrated in broccoli (Forney et al., 2003), respiration rate increases when the dose of ozone being used is too high which would be due to the damage caused to the produce.

Ozone exposure at 0.1 $\mu\text{mol mol}^{-1}$ had no significant effect on AsA and DHA content in red bell peppers (Table 4) when compared with control samples, whereas when applied at 0.3 $\mu\text{mol mol}^{-1}$, the AsA content after 14 d of storage at 14 °C was reduced. Even though vitamin C content after 7 d of storage was significantly enhanced in peppers exposed to ozone at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$ (Table 4), this effect was not maintained in the end of the storage period due to increase in AsA content in control samples. A number of

authors have found no effect of ozone exposure on AsA content in whole tomatoes cyclically exposed to gaseous ozone at 4 $\mu\text{mol mol}^{-1}$ for 30 min every 3 h (Aguayo et al., 2006) or continuously exposed to ozone at 1 $\mu\text{mol mol}^{-1}$ for 6 d (Tzortzakis et al., 2007a). Zhang et al. (2005), on the other hand, observed that in celery washed with ozonated water at 0.03-0.18 $\mu\text{mol mol}^{-1}$ for 5 min, changes in vitamin C content in response to ozone were dose-dependent. In their study, vitamin C content was significantly increased at all doses of ozone after 3 and 6 d of storage. Even though, vitamin C content after 9 d of storage in celery treated at the highest dose (0.18 $\mu\text{mol mol}^{-1}$) was significantly reduced when compared with their counterparts treated at 0.03 and 0.08 $\mu\text{mol mol}^{-1}$, it was not significantly different from control. Changes in AsA content are not surprising as AsA is a key antioxidant in plant tissue (Conklin, 2001; Mittler, 2002) and its role is to scavenge ROS that are produced in excess under stress conditions, e.g. high dose of ozone. The findings from this study imply that there is a threshold, where at certain dose of ozone, i.e. high concentration or prolonged exposure, nutritional quality of peppers may be reduced.

Table 4

Effect of ozone exposure at 0.1 and 0.3 $\mu\text{mol mol}^{-1}$ on changes in antioxidant content of red bell peppers during the storage for 14 d at 14 °C. Data represent mean values from 12 replicates. Different letters indicate that values are significantly different ($P < 0.05$).

Time of storage	Treatment	AsA (mg/100 g FW)	DHA (mg/100 g FW)	vitamin C (mg/100 g FW)	Total phenolics (mg GAE/100 g FW)
Day 7	ctrl	60.9 b	4.5 a	65.4 b	79.7 ab
	0.1 $\mu\text{mol mol}^{-1}$	67.5 ab	4.0 a	71.5 a	88.4 a
	0.3 $\mu\text{mol mol}^{-1}$	66.5 ab	6.6 a	73.1 a	74.3 b
Day 14	ctrl	75.8 a	1.9 a	77.7 a	87.8 ab
	0.1 $\mu\text{mol mol}^{-1}$	69.8 a	1.1 a	70.9 a	92.3 a
	0.3 $\mu\text{mol mol}^{-1}$	64.8 b	2.2 a	67.0 b	81.6 ab

The content of total phenolics was increased in red bell peppers exposed to ozone at 0.1 $\mu\text{mol mol}^{-1}$ (Table 4), while it was not affected in their counterparts exposed to ozone at 0.3 $\mu\text{mol mol}^{-1}$. Total phenolics content has previously been found to be increased in ozone treated banana (Allothman et al., 2010), kiwi (Minas et al., 2012), papaya (Ali et al., 2014) and pineapple (Allothman et al., 2010) when compared with control fruit stored in clean air. Increase in phenolics was also reported (Yeoh et al., 2014) when papaya was treated with gaseous ozone at 9.2 $\mu\text{mol mol}^{-1}$ for 10 or 20 min prior to storage, but not when the exposure time (30 min) was too long. This may be explained by antioxidant capacity of phenolic compounds, i.e. if the dose of ozone is too high it could result in excess oxidative stress and production of ROS which then need to be scavenged by

antioxidants, e.g. phenolic compounds. These findings suggest that similarly to changes in AsA, the content of phenolic compounds is affected in response to ozone in a dose-dependent manner. The possible mechanism may be associated with increased activity of phenylalanine ammonia lyase (PAL; EC 4.3.1.5) in ozone treated samples or reduced activity of polyphenoloxidase (PPO; EC 1.14.18.1) and/or peroxidase (POD; EC 1.11.1.7), which are all involved in polyphenol biochemistry (Toivonen and Brummell, 2008).

In the case of cucumbers and zucchini, ozone exposure at $0.1 \mu\text{mol mol}^{-1}$ had no effect on chemical composition (ascorbic acid and total phenolics) of the fruit (data not presented), which may be related to its limited penetration through the cuticle. No changes in AsA and phenolics may also suggest that the level of ozone at $0.1 \mu\text{mol mol}^{-1}$ is too low to induce production of reactive oxygen species (ROS). On the other hand, since ozone cannot penetrate deeply into the fruit, it could have induced some biochemical/physiological changes in the skin by interacting with cell wall constituents, as has previously been reported (An et al., 2007).

3.5. Disease incidence

In trial 1 with red bell peppers, no sign of rotting was observed on the fruit surface during the 14 d of storage at 14°C . Once the fruit were cut into halves, the fungal infection was noticed in 4-8% of the fruit, regardless of the treatment, on the septum and placental region probably due to latent infection present in the fruit at harvest. Since ozone cannot penetrate deeply into the fruit, fungal development could not be prevented. In trial 2 with red bell peppers, no sign of rotting was observed on the fruit surface after 7 d of storage. After 14 d, however, no sign of rotting was observed only in peppers continuously exposed to ozone at $0.3 \mu\text{mol mol}^{-1}$. The growth of fungi on stem and peduncle was observed in 8.3% of the fruit continuously exposed to ozone at $0.1 \mu\text{mol mol}^{-1}$, while in 25% of the control peppers. In contrast to trial 1, no fungal infection was noticed inside the fruit used in the trial 2, i.e. the latent infection was not present. The finding from this study suggests that continuous exposure to ozone at doses of 0.1 and $0.3 \mu\text{mol mol}^{-1}$ may efficiently reduce disease incidence in whole red peppers.

In cucumbers, no sign of rotting was observed on the fruit surface after 6 d of storage at 12°C . After 10, 13 and 17 d of storage, however, disease incidence was found on 8.3, 16.6, 33.3% and 8.3, 8.3, and 16.6% of the control and ozone-exposed fruit, respectively. In zucchini, on the other hand, fungal development was noticed after 10 d of storage at 8°C on 8.3% of the fruit, regardless the treatment. After 13 and 17 d, signs of rotting were observed on 16.6% of both control and treated fruit. Fungal development on cucumbers and zucchini was not prevented by continuous exposure to ozone at $0.1 \mu\text{mol mol}^{-1}$.

Continuous exposure to ozone at $0.1 \mu\text{mol mol}^{-1}$ reduced disease incidence in red bell peppers but not in cucumbers and zucchini. This difference may be due to differences in the initial microbial counts, which were not determined in this study or due to several other factors, including skin characteristics of the produce and its sensitivity to ozone, which varies between different commodities.

3.6. Effect on ethylene

In the trials with red bell peppers, ethylene concentration within the containers was within the range of $0.005\text{-}0.040 \mu\text{mol mol}^{-1}$ in containers with control samples and those continuously exposed to ozone at $0.1 \mu\text{mol mol}^{-1}$. The concentration of ethylene was only slightly but not significantly reduced, below $0.010 \mu\text{mol mol}^{-1}$, in containers with peppers stored with ozone at $0.3 \mu\text{mol mol}^{-1}$. In the trials with cucumbers and zucchini, the concentrations of ethylene in the containers were even lower, i.e. in the range of $0.002\text{-}0.014 \mu\text{mol mol}^{-1}$ in cucumbers and $0.010\text{-}0.025 \mu\text{mol mol}^{-1}$ in zucchini, respectively. Continuous exposure to ozone at $0.1 \mu\text{mol mol}^{-1}$ had no measureable effect on ethylene concentration within the containers. The lack of effect of ozone on the ethylene concentration could be explained by low ethylene concentrations in all trials (not exceeding $0.040 \mu\text{mol mol}^{-1}$), thus making it difficult to observe any reduction. The other reason could be that the removal of ethylene at ozone doses of 0.1 and $0.3 \mu\text{mol mol}^{-1}$ is rather slow. Palou et al. (2001) reported that in an empty 59.78 m^3 container after continuous supply of ozone for 24 h, resulting in an ozone concentration of $0.3 \mu\text{mol mol}^{-1}$, the concentration of ethylene was only reduced by 1% of its initial value. The efficiency of ozone in ethylene removal, however, was found (Palou et al., 2001) to be significantly increased at higher doses of ozone, i.e. continuous ozone supply at above $1 \mu\text{mol mol}^{-1}$.

4. Conclusion

Continuous exposure to ozone at $0.1 \mu\text{mol mol}^{-1}$ enhanced fructose and glucose, and total phenolics content in red bell peppers. At the same time, it had no significant effect on weight loss, texture and colour. In the case of cucumbers and zucchini, continuous exposure to ozone at $0.1 \mu\text{mol mol}^{-1}$ reduced weight loss and improved texture maintenance, while having no significant effect on their biochemistry. Ozone cannot penetrate deeply into the fruit, thus it must have induced biochemical/physiological changes in their skin. It is apparent from this work that continuous exposure to ozone at $0.1 \mu\text{mol mol}^{-1}$ had no adverse effect on quality of red peppers, cucumbers and zucchini. The findings from this study suggest that when used at the proper dose, not too high to cause the damage or nutritional quality loss, i.e. continuous exposure to ozone at $0.1 \mu\text{mol mol}^{-1}$ is a promising method for shelf-life extension of cucumbers and zucchini. Effects of ozone are not considered to be associated with removal of ethylene, as ethylene concentrations were lower than $0.04 \mu\text{mol mol}^{-1}$ in all cases,

with no significant effect of ozone treatment. Further work, however, is still needed to better understand the exact mechanism of ozone action in case of red peppers and whether its use has the potential for industrial use.

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